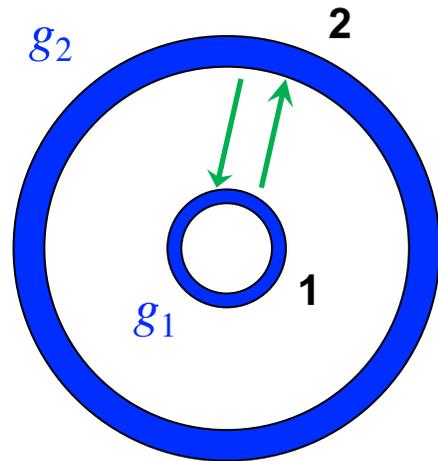


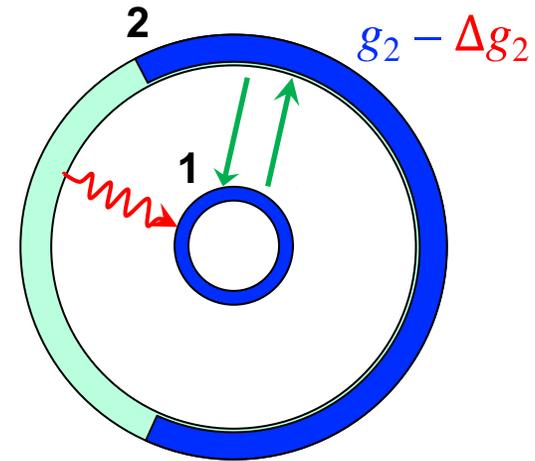
Nonequilibrium Thermodynamics Under Collisional-Radiative Equilibrium

Isolated two-level atom



$$\frac{N_2}{N_1} = \frac{g_2}{g_1} e^{-\epsilon_{21}/kT}$$

Reduce occupation probability by radiative decay



$$\frac{N_2}{N_1} = \frac{g_2 - \Delta g_2}{g_1} e^{-\epsilon_{21}/kT}$$

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We have applied an atomic occupation-probability approach to thermodynamic energy and pressure of a radiating gas under collisional-radiative equilibrium



- **Collisional-radiative equilibrium (CRE) is the very important steady state of continuous, freely, and irreversibly escaping radiation**
- **Properties of matter under CRE are functions of local thermodynamic variables, so they can be tabulated for use in numerical radiation-hydrodynamic simulations with the tools of classical equilibrium thermodynamics**
- **The simplest atomic system shows small reductions in pressure and energy that are characteristic of radiation effects under CRE that are absent under local thermodynamic equilibrium (LTE)**

Collaborators



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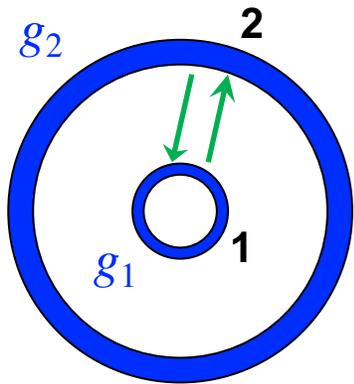
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The Hummer–Mihalas* framework introduces nonideal equation-of-state (EOS) effects through modified atomic state occupation probabilities

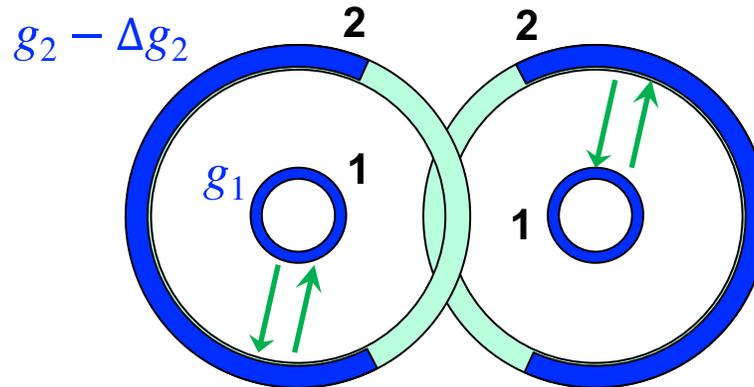
Isolated two-level atom
Collisional LTE \rightleftharpoons



$$\frac{N_2}{N_1} = \frac{g_2}{g_1} e^{-\epsilon_{21}/kT}$$

Add **radiative decay** to collisional equilibrium:

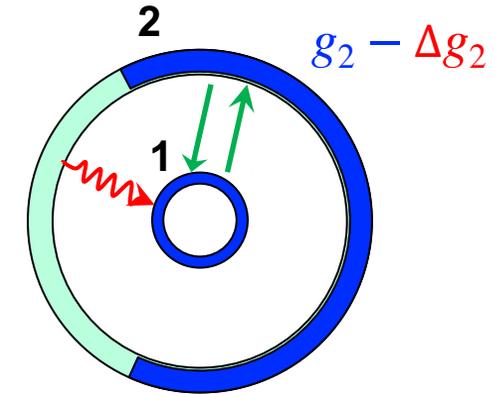
Reduce occupation probability by pressure ionization



$$\frac{N_2}{N_1} = \frac{g_2 - \Delta g_2}{g_1} e^{-\epsilon_{21}/kT}$$

$$n_e C_{12} N_1 = (n_e C_{21} + A_{21}) N_2$$

Reduce occupation probability by **radiative decay**



$$\frac{N_2}{N_1} = \frac{g_2 - \Delta g_2}{g_1} e^{-\epsilon_{21}/kT}$$

$$\frac{N_2}{N_1} = \frac{g_2}{g_1} \frac{e^{-\epsilon_{21}/kT}}{1 + A_{21}/n_e C_{21}}$$

* D. G. Hummer and D. Mihalas, *Astrophys. J.* **331**, 794 (1988).

The need for non-LTE models of radiating matter in rad-hydro numerical simulations has motivated many simplifying approximations

- Busquet (RADIOM, 1993):* Obtain non-LTE ionization and opacity from LTE tables using an “ionization temperature”

$$Z_{\text{non-LTE}} = Z_{\text{LTE}}(T_{\text{non-LTE}}, V)$$



Epstein (1998):** Calculate and tabulate correct ionization and opacity tables in CRE as an important option to LTE tables

- Busquet (1982):† Modify LTE excitation and ionization ratios for radiative decay using phenomenological factors



This simple CRE correction factor form is familiar and nearly 40 years old

$$\frac{N_2}{N_1} = \frac{g_2}{g_1} e^{-\varepsilon_{21}/kT} \left/ (1 + A_{21}/n_e C_{21}) \right. \approx \frac{g_2}{g_1} e^{-\varepsilon_{21}/kT} \left/ (1 + \alpha \varepsilon_{21}^3 T^{1/2} V / N_e) \right.$$

- Zimmerman & More (1980),‡ Hummer & Mihalas (1988):‡‡ Formulate nonideal effects (e.g., pressure ionization) in the free energy by modifying the occupation probabilities of atomic states



A CRE free energy based on the same CRE atomic model used for opacity and emissivity will assure overall self-consistency and a thermodynamic consistent EOS.

$$\frac{N_i}{N} = g_i e^{-[\varepsilon_i - (\partial f / \partial N_i)] / kT} / \tilde{Z}^I$$

* M. Busquet, Phys. Fluids B **5**, 4191 (1993).

** R. Epstein *et al.*, Bull. Am. Phys. Soc. **43**, 1666 (1998).

† M. Busquet, Phys. Rev. A **25**, 2302 (1982).

‡ G. B. Zimmerman and R. M. More, J. Quant. Spectrosc. Radiat. Transf. **23**, 517 (1980).

‡‡ D. G. Hummer and D. Mihalas, Astrophys. J. **331**, 794 (1988).

Local thermodynamic equilibrium thermodynamics must be modified to describe matter in collisional-radiative equilibrium

- Pressure and internal energy obtained from the free energy $F = E - TS$ are thermodynamically consistent
- In LTE, constituent populations are constrained by chemical equilibrium
- In CRE, constituent populations are not in statistical equilibrium
- A CRE chemical equilibrium with modified chemical potentials is needed to restore thermodynamics, EOS tables, etc., valid for rad-hydro simulations of matter under CRE conditions

$$P = -\left(\frac{\partial F}{\partial V}\right)_{T,\{N_i\}} \quad E = -T^2 \left(\frac{\partial (F/T)}{\partial T}\right)_{V,\{N_i\}}$$

$$\sum_i \mu_i dN_i = 0 \quad N_i^{\text{LTE}}(T, V)$$

$$\sum_i \mu_i dN_i \neq 0 \quad N_i^{\text{CRE}}(T, V) \neq N_i^{\text{LTE}}(T, V)$$

$$\sum_i \mu_i^{\text{CRE}} dN_i = 0$$

• Boltzmann two-level atom:
$$\frac{N_2}{N_1} = \frac{\frac{g_2}{g_1} e^{-\varepsilon_{21}/kT}}{1 + A_{21}/n_e C_{21}} = \frac{g_2}{g_1} e^{-(\varepsilon_{21} - \delta\mu_2^{\text{CRE}})/kT}$$

• Saha ionization:
$$\frac{N_{j+1} n_e}{N_j} = \frac{2g_{j+1} \left(\frac{2\pi m_e kT}{h^2}\right) e^{-\chi_j/kT}}{1 + R_{j+1,j}/n_e C_{j+1,j}}$$

The full Hummer–Mihalas* free energy can be applied to any composition and atomic model that is formulated in terms of state populations

- Correct the free energy for state occupation probability changes caused by radiative decay

$$F_{\text{CRE}} = F_{\text{LTE}} + f(T, V, \{N_i\}) \quad \frac{N_i}{N} = g_i e^{-[\varepsilon_i - (\partial f / \partial N_i)] / kT} / \tilde{Z}^I$$

subject to the constraint $\sum_i (\partial F_{\text{CRE}} / \partial N_i) dN_i = 0$, where $\tilde{Z}^I = \sum_i g_i e^{-[\varepsilon_i - (\partial f / \partial N_i)] / kT}$

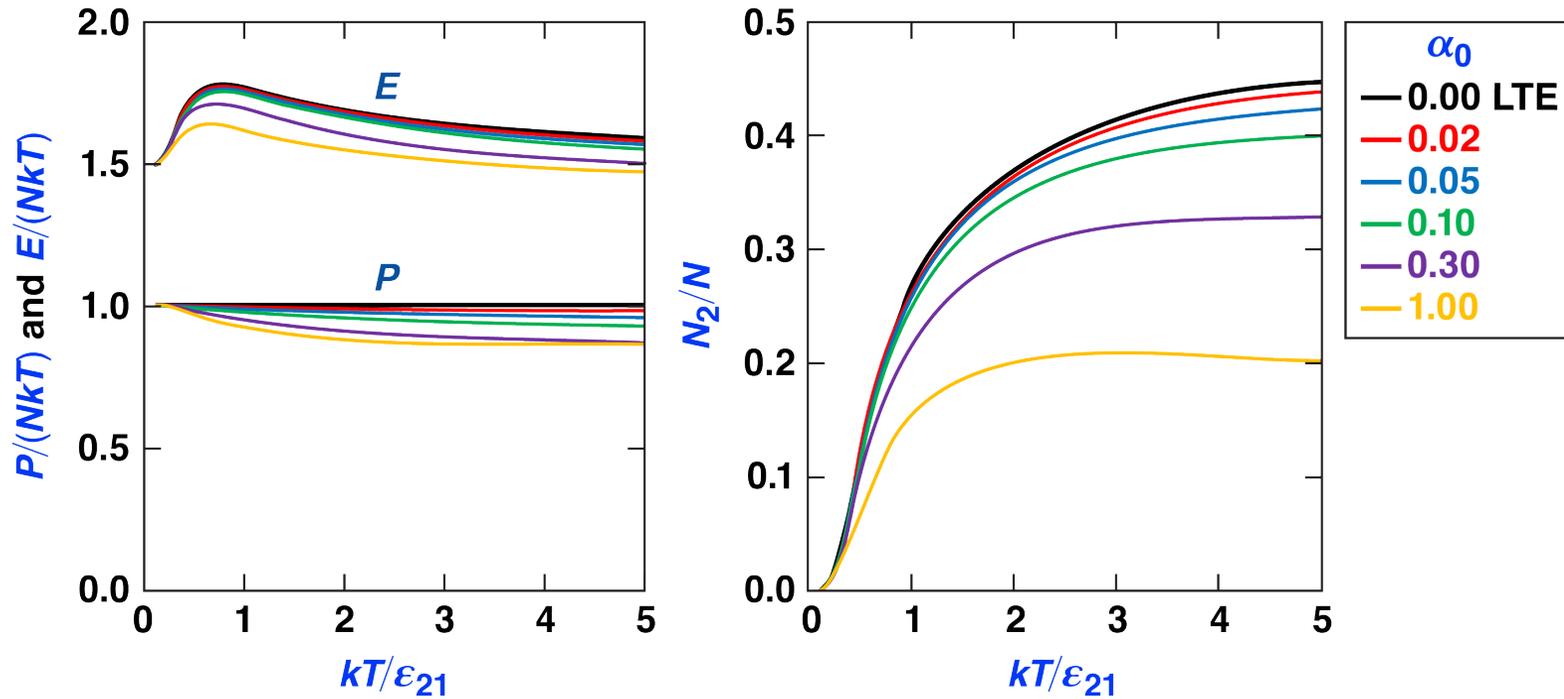
- Can we find the set of CRE-modified chemical potentials equivalent to a general CRE solution?

$$\partial f / \partial N_i = -\delta\mu_i \approx kT N_i \ln(1 + \beta_i T^\nu V^\eta)$$

- Simultaneous LTE and CRE calculations give $N_i^{\text{CRE}} / N_i^{\text{LTE}}$ ratios as occupation probability modification values
- A complete set of partial derivatives $\{\partial\mu_i / \partial T|_V, \partial\mu_i / \partial V|_T\}$ at each tabulation point is needed

* D. G. Hummer and D. Mihalas, *Astrophys. J.* **331**, 794 (1988).

In the two-level atomic gas, nonideal CRE EOS effects arise from volume-dependent radiative free-energy terms



$$A_{21}/n_e C_{21} = \alpha_0 \left(\frac{kT}{\epsilon_{21}} \right)^{1/2} \left(\frac{V}{V_0} \right)$$

- Ideal gas behavior is recovered in the non-radiative limit
- This CRE nonideal behavior resembles the effect of the **volume-dependent interaction energy** in a Van der Waals gas

$$P = kT \left\{ N - N_2 V \frac{\partial \ln \left[1 + \alpha_0 \left(\frac{kT}{\epsilon_{21}} \right)^{1/2} \left(\frac{V}{V_0} \right) \right]}{\partial V} \right\}_T$$

$$E = \frac{3}{2} NkT + N_2 \left[\epsilon_{21} - kT^2 \frac{\partial \ln \left(1 + A_{21}/n_e C_{21} \right)}{\partial T} \right]_V$$

$$P = \frac{NkT}{V - Nb} - \frac{N^2 a}{V^2}$$

The rate ratio $A_{21}/n_e C_{21}$ that indicates where CRE affects EOS also indicates generally where CRE is applicable.

We have applied an atomic occupation-probability approach to thermodynamic energy and pressure of a radiating gas under collisional-radiative equilibrium



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- **The simplest atomic system shows small reductions in pressure and energy that are characteristic of radiation effects under CRE that are absent under LTE**