

**Investigation of the effect of a plastic ring on the uniformity of ice layers  
in cryogenic inertial confinement fusion targets**

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**Abstract:**

A program has been written to model the evolution of the cryogenic ice layer during the layering of a target for inertial confinement fusion. The program simulates the thermal diffusion in the layering sphere, as well as the sublimation and deposition of the deuterium. This will facilitate the design and production of more-uniform targets. It has been found that any nonuniformities caused by the plastic rings of the recently proposed “Saturn” targets are too small to significantly impact fusion performance.

**1. Introduction**

Controlled nuclear fusion is a promising technology for producing clean, abundant energy. Laser-based inertial confinement fusion (ICF) is one method under investigation for producing the extreme temperatures and pressures required for fusion reactions to occur. In direct-drive ICF, an array of powerful laser beams fires upon a target from many directions, heating and compressing the target<sup>1</sup>. The target is a spherical plastic shell containing the hydrogen isotopes deuterium ( $D_2$ ) and tritium ( $T_2$ ). When the target is compressed, the deuterium and tritium nuclei fuse to form helium nuclei and energetic neutrons. (Another approach, indirect-drive ICF<sup>2</sup>, involves laser beams entering the target chamber from the top and bottom and irradiating an x-ray-producing target. These x-rays then “indirectly” drive the spherical fuel-containing target.) Targets used in current laser-based ICF experiments generally contain only deuterium, as tritium is radioactive and hard to work with.

Cryogenic targets are spherical plastic shells with thin layers of solid deuterium and tritium on the inner surface of the shell. By using solid fuel, cryogenic targets can store much more fuel than traditional, gas-filled targets, increasing the likelihood of a

chain reaction. The plastic shells are first filled with gaseous deuterium and tritium, which are transformed to solid when the shells are cooled down. The target is then slowly heated in a device called a layering sphere to generate a more even ice layer, which improves the fusion performance.

Unfortunately, the heating inside the layering sphere is not perfectly uniform. For example, the so-called “Saturn” targets<sup>3</sup> for polar direct drive at the National Ignition Facility have small plastic rings around the equators, which can cause uneven heat conduction in the layering sphere. This can introduce nonuniformities in the ice layer, reducing the fusion performance. As the “Saturn” target has only been proposed very recently, it is not clear how severe the ice nonuniformity would be and whether it would have a significant impact on fusion performance. A program has been written to model the evolution of the cryogenic ice layer during the layering process, including simulation of the thermal diffusion in the layering sphere as well as the sublimation and deposition of the deuterium. The program was initially intended to answer the “Saturn” ring target question, but it is also flexible enough to model other target designs. It is believed to be the first program of its type.

### **1.1 Cryogenic targets**

Cryogenic targets currently being used on the OMEGA laser system<sup>4</sup> are about 900  $\mu\text{m}$  in diameter and contain solid deuterium. A cryogenic target typically has a 3  $\mu\text{m}$ -thick plastic shell, with a 100  $\mu\text{m}$ -thick deuterium ice shell on the inner surface of the plastic shell. The center contains deuterium vapor (see figure 1).

### **1.2. Cryogenic target formation**

The formation of the cryogenic target involves several steps. First, a spherical

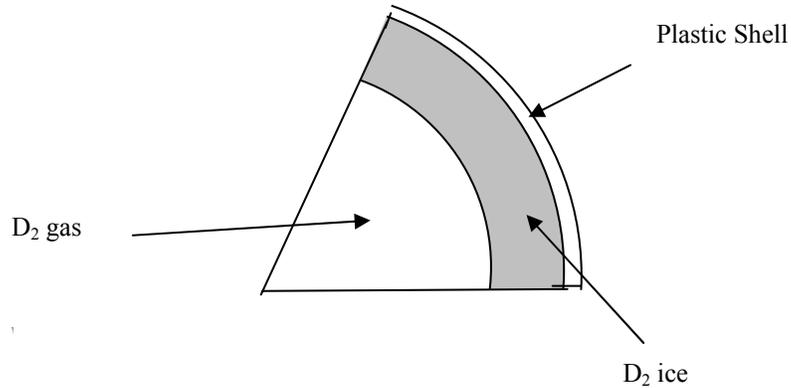


Figure 1: Cryogenic target cross section

plastic shell is filled, through diffusion, with deuterium gas to a pressure of about 1000 atm. The target is then cooled below the triple point of  $D_2$ , first entering the liquid phase, then freezing into a layer of solid  $D_2$ . The  $D_2$  layer is not uniform, as gravity pulls the  $D_2$  to the bottom of the shell when it is in the liquid phase. Then, the target is placed in a layering sphere (figure 2), a spherical cavity with reflective, gold-coated walls<sup>5</sup>. An optical fiber is used to introduce low-power infrared light into the layering sphere. The light reflects off the walls and is absorbed by the  $D_2$  ice. The rough surface of the wall ensures that the infrared light is scattered in all directions. In addition, the infrared light may bounce back and forth several times before it hits the target. Consequently, the target is heated uniformly in all directions. The thicker areas of the ice absorb more light and become hotter than the thinner areas. The ice in the thicker areas sublimates faster than the ice in the thinner, cooler areas, and the resulting gas

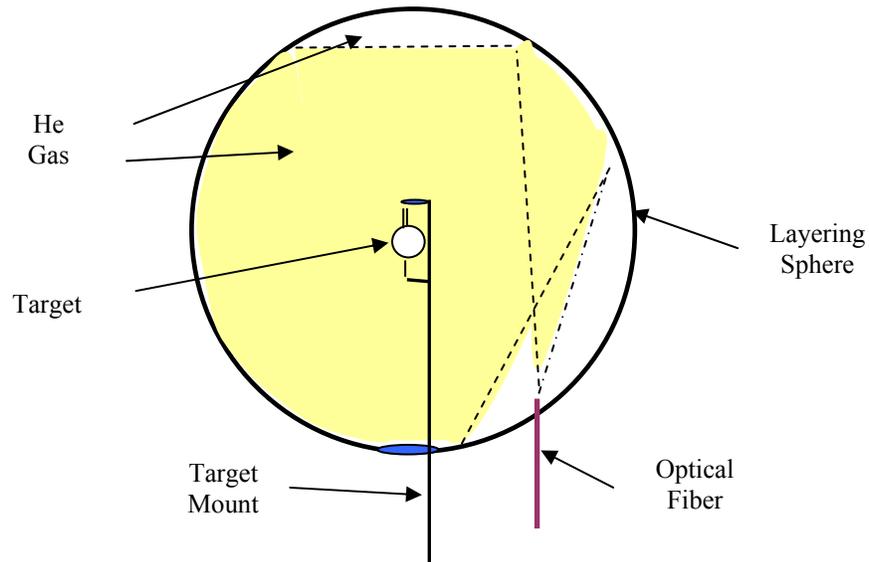


Figure 2: Layering sphere. The target is supported by a target mount. The optical fiber introduces the infrared light (yellow area), which is scattered from the wall and heats the target uniformly. The helium gas in the layering sphere conducts the heat from the target to the wall.

preferentially condenses on the cooler, thinner areas (figure 3). The heat is removed from the target by conduction through the helium gas to the wall of the layering sphere, which is kept at a constant temperature of 18.6 K. The layering process takes about 24 hours,

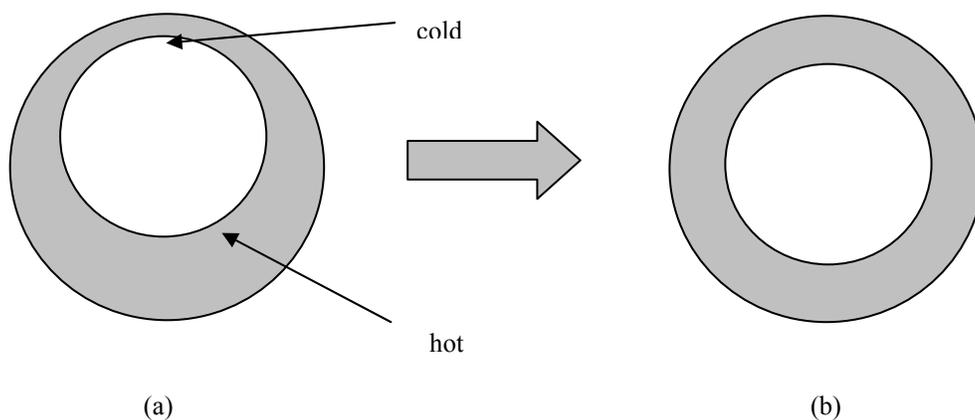


Figure 3: A cryogenic target (a) before and (b) after the layering process. The thicker portions become hotter due to heating by infrared light, so deuterium sublimates from the thick portions and deposits on the thin portions, resulting in a more even ice layer.

and results in an isotherm along the inner surface of the ice layer. An even ice layer is required because an uneven ice layer results in uneven compression and poor fusion performance. A nonuniformity of 1% can be tolerated in the ice layer. In a 100  $\mu\text{m}$  ice layer, this translates to a nonuniformity of 1  $\mu\text{m}$ .

### 1.3 Saturn Targets

The National Ignition Facility (NIF) is a facility for laser-based ICF research scheduled for completion in 2010. The new, 192-beam laser is designed to deposit 1.8 MJ of energy on a target 3 mm in diameter. The greater amount of fuel in the larger targets increases the probability of an energy gain, in which the fusion energy generated is greater than the input laser energy. However, the NIF is designed to perform indirect-drive laser-based ICF experiments and has no laser beams aimed at the equator of the target. To perform direct-drive experiments, a novel target design has been proposed.

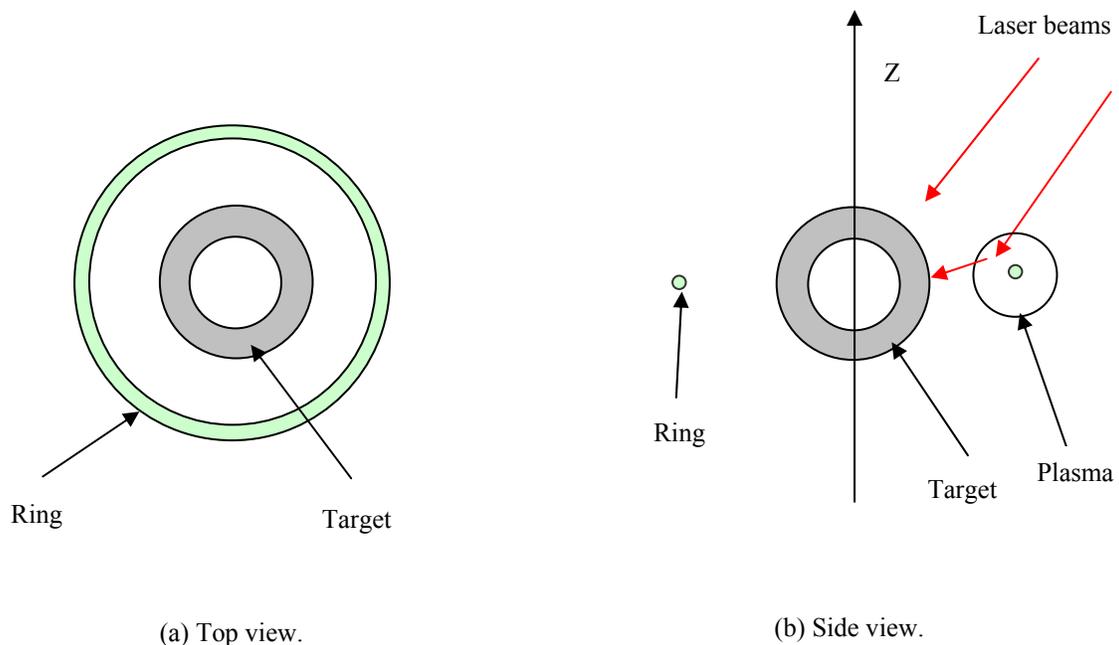


Figure 4: "Saturn" target.

The new target, the so-called “Saturn target,” has a small plastic ring around the equator (figure 4)<sup>3</sup>. When the laser is fired, the plastic ring forms a plasma, which causes some of the laser beams to refract, striking the target at the equator.

The Saturn target raises uncertainty in layering. The ring has a higher thermal conductivity than the helium gas in the layering sphere. Hence, more heat is conducted from the equatorial regions of the target than from the polar regions. This causes the ice in the equatorial regions to be thicker than that in the polar regions. However, the degree to which the ring affects the ice thickness is not known.

The primary goal of this project is to ascertain the effect of the ring on the ice layer and, in particular, to determine the size of the ice perturbation.

## **2. The Program**

The project entailed writing a program to model the formation of the ice layer in a cryogenic target. The user defines the target structure, an initial ice layer (specifically the radius of the inner ice surface for each direction ( $\theta$ ,  $\phi$ ) with respect to the vertical axis), and an initial temperature for each part of the system. The infrared heating of the target and the sublimation and condensation of the ice layer are simulated over a period of time defined by the user. The program displays the result with plots showing the new location of the ice layer, as well as the temperature contours inside the target.

The entire layering sphere, including the cryogenic target, is divided into three-dimensional (3-D) grid cells, of which 2-D cross sections are shown in figure 5. This is done in spherical geometry, along the  $r$ ,  $\theta$ , and  $\phi$  dimensions. The grid is typically chosen to be denser in the ice layer, especially near the interface between the ice and gas.

(The fine zoning is expanded beyond the ice layer to accommodate simulations involving severe ice nonuniformities.)

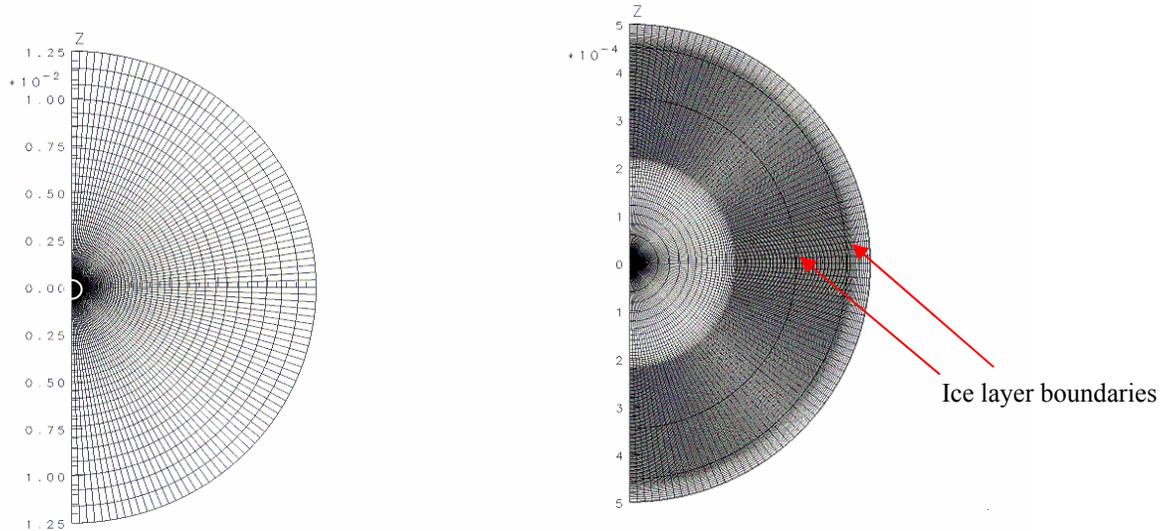


Figure 5: (a) The entire layering sphere divided into grid cells (outer radius=1.25 cm), and (b) the area outlined in white in greater detail (radius = 500  $\mu\text{m}$ ). Note the high density of grid cells in and near the ice layer, whose boundaries are shown with dark lines.

The temperature of each cell is stored in a 3-D array, as are the thermal conductivity, the specific heat, and the amount of heat each cell receives from the infrared laser. A two-dimensional array, indexed in the  $\theta$  and  $\phi$  directions, keeps track of the radius of the ice-gas interface. Although this particular problem (Saturn ring) is 2-D in nature due to the symmetry in  $\phi$ , the program was written in 3-D to allow for more complex future applications. The program performs a large number of time steps, each of which is typically a small fraction of a second. During each time step, the new temperature of each cell is calculated, taking into account the infrared heating and thermal conduction. The amount of sublimation and condensation at each interface cell is also calculated, and the new radius of the interface is determined.

The rate of sublimation is a function of the temperature of the solid. Although it is not a linear function, we are working with a very small temperature range near equilibrium. Thus,  $\frac{dN_s}{dt}$ , the rate of sublimation per unit area in molecules/m<sup>2</sup>s, is linearized as:

$$\frac{dN_s}{dt} = N_0 + \alpha(T_{solid} - T_{ref}), \quad (1)$$

where  $T_{ref}$  is the reference temperature of 18.6 K,  $T_{solid}$  is the temperature of the deuterium ice,  $N_0$  is the rate of sublimation at the reference temperature, and  $\alpha$  is a constant coefficient. (See the appendix for numerical values of  $\alpha$  and other physical constants.)

Similarly,  $\frac{dN_c}{dt}$ , the rate of deposition is linearized as:

$$\frac{dN_c}{dt} = n_{gas} [\beta - \gamma(T_{gas} - T_{ref})], \quad (2)$$

where  $T_{gas}$  is the temperature of the gas,  $n_{gas}$  is the density of the gas in molecules per cubic meter, and  $\beta$  and  $\gamma$  are constant coefficients. Subtracting (1) from (2) gives the overall rate of deposition of molecules per unit area:

$$\frac{dN}{dt} = n_{gas} [\beta - \gamma(T_{gas} - T_{ref})] - N_0 - \alpha(T_{solid} - T_{ref}), \quad (3)$$

In addition, the thermodynamic activity is modeled by the following:

$$\frac{\partial}{\partial t} C_v T = \nabla \cdot \kappa \nabla T + Q_{IR} + Q_L, \quad (4)$$

where  $C_v$  is the specific heat,  $T$  is the temperature,  $\kappa$  is the conductivity,  $Q_{IR}$  is the infrared heat, and  $Q_L$  is the latent heat per unit volume, which is proportional to the rate of deposition of molecules per unit area. It seems reasonable to deposit the latent heat in

the solid (rather than the gas) as the slow molecules in the thermal distribution in the gas will excite lattice vibrations in the solid lattice when they are captured. Some of this energy may return to the gas through thermal conduction.

The latent heat is believed to be deposited over a very thin layer, but its distribution is difficult to determine. However, since the thermal conduction will rapidly remove the steep gradients, the exact distribution is not very important here. Numerically, we deposit all the latent heat in the first boundary cell that contains ice. This cell will have

$$Q_L = \frac{L}{\Delta x} \frac{dN}{dt}, \quad (5)$$

where  $L$  is the latent heat per molecule and  $\Delta x$  is the ice thickness in the cell.

The sublimation and heat transfer occur simultaneously. To calculate the temperature changes, the amount of sublimation that occurred and thus  $Q_L$  must be known. Also, the temperatures of the ice and gas at the interface must be known to determine the amount of sublimation that occurs. However, to solve the sublimation and heat transfer in one step would be difficult and computationally intensive, so it was decided to decompose them into two separate procedures. We have separated the problem into a linear equation (Equation (4)) and a few linear and nonlinear function evaluations (Equations (3) and (5)). The heat transfer equation has been well studied. Many mature algorithms exist that can efficiently solve this equation. Equations (3) and (5) can be evaluated in a straightforward manner. As the time step is small the error introduced due to the separation of heat transfer and sublimation/condensation is negligible.

The ice radii calculated after each time step typically do not fall on a cell border. As a result, the interface cell contains both ice and gas (see figure 6). The cells across the

ice-gas interface are modeled using a linear mixture of parameters from both ice and gas, according to the ice/gas ratio within the cell. Specifically, once the ice radii are evaluated, the cells are searched to determine which ones contain the interface. The gas and ice volumes within the interface cells are calculated. The specific heat and the conductivity of the cell are then computed as

$$C_v = pC_{vGas} + (1-p)C_{vSolid} , \quad (6)$$

and

$$\kappa = p\kappa_{Gas} + (1-p)\kappa_{Solid} , \quad (7)$$

where  $p$  and  $(1-p)$  are the proportions (in volume) of gas and solid contained in the cell, respectively.

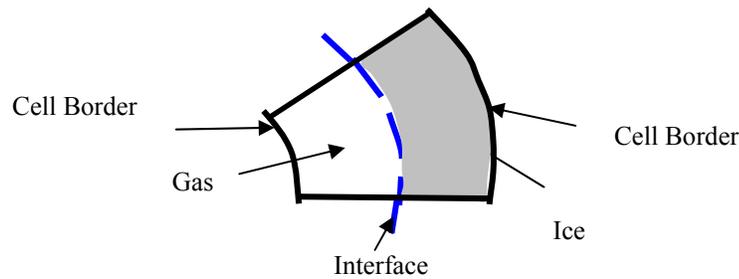


Figure 6: Cell containing the gas/ice interface. Both ice and gas are present in the cell.

A natural approach to predicting the ice shape at the end of the layering process is to set the initial condition to be as representative of the real situation as possible, and run the simulation for the same duration as the real process. As the software time step is typically a few hundred microseconds and the layering requires 24 hours, the number of time steps required is in the hundreds of millions. Two issues emerged with this huge number. First, it would take days to perform one simulation, even with a modern computer. Second, computation errors, such as rounding errors, accumulate during each

step of the calculation. The total error, which is proportional to the number of steps, may be large enough to make the calculation meaningless. The introduction of variable time step techniques, which will be discussed later, can reduce the number of time steps by a factor of about five. However, the resulting number of steps is still extremely large.

It was decided to try a different method. When the layering process approaches the end, the system typically enters a steady state, which means that each part of the system remains almost unchanged. In our application, steady state implies that the sublimation is balanced by condensation, and the temperature of each cell remains constant. The steady state is determined by the system configuration including boundary conditions, heat sources, etc. It is very insensitive to a wide range of initial conditions. Consequently, we may calculate the steady state solution with a small amount of steps if we can find an initial condition (including initial ice geometry and initial temperature) that is close to the steady state. It turns out that this is not a very difficult task. It is known that the steady state for the target without the Saturn ring is a uniform shell. The introduction of the ring does change the steady state. Nevertheless, the magnitude of the change is relatively small. As a result, it can be reasonably assumed that using a uniform shell as the starting point, the simulation may reach the steady state in a short period of time.

Another method of reducing the number of time steps involves applying variable time steps. Specifically, during the initial transition period when the system drastically changes, a smaller time step is applied. Afterwards, when the system is near the equilibrium temperature, we take relatively large time steps. This variable step size scheme was developed based on several observations. A large step size saves time steps,

but it may cause significant calculation error and even instability. The step-by-step calculation of equations (3)-(5) assumes that within each time step the system variables including ice radius and cell temperature do not change very much and thus can be approximated as constants and linear functions. The above assumption could be violated once a large time step is adopted. However, we noticed that the error in approximation does not only depend on the time step size, but also on how quickly the system changes. When the system remains steady, even a relatively large step does not introduce noticeable error. As a result, the variable time step size approach serves both our goals: fewer time steps and acceptable calculation error. The maximum ice thickness increment is used as an indicator for the system change. The time step size is determined to be inversely proportional to the indicator. A ceiling is imposed to prevent too large a step size when the change on the previous step approaches zero. More specifically, the step size for step  $i$  is determined as

$$\Delta t(i) = \text{Min}\left(\frac{\Delta r_d \Delta t(i-1)}{\text{Max}_{\theta\varphi} |\Delta r_{\theta\varphi}(i-1)|}, \Delta t_{\text{max}}\right), \quad (8)$$

where  $\Delta t_{\text{max}}$  is the maximum step size allowed,  $\Delta r_d$  is the desired radius change, typically around 10  $\mu\text{m}$ .  $\Delta r_{\theta\varphi}(i-1)$  is the ice radius change in the direction of  $(\theta, \varphi)$  that occurred in the previous time step,  $\Delta t(i-1)$  is the size of the previous time step, and the maximization is over all directions. This variable step size approach results in a reduction of the number of the steps by a factor of about five.

### 3. Results

The ice shape of the layered ‘‘Saturn’’ target was determined using the program. As discussed in Section 2, the initial condition was set with an ice layer of uniform

thickness. The heat absorption of the ice was set at about  $46 \text{ kW/m}^3$ , approximately the same as the heat produced through beta decay of deuterium-tritium ice<sup>6,7</sup>. The target has a diameter of  $900 \text{ }\mu\text{m}$ , an ice layer thickness of  $100 \text{ }\mu\text{m}$ , a ring cross section diameter of  $300 \text{ }\mu\text{m}$ , and a ring major radius (distance from the center of the ring to the origin) of  $1100 \text{ }\mu\text{m}$ . The results are summarized in Figures 7 through 9.

Figure 7 shows temperature contours for the target, (a) at time  $t = 5$  seconds and (b) after an hour, plotted in polar coordinates. The lines in the contour maps are isotherms, representing locations with the same temperatures. The ice surface radii, which are not plotted, are about  $347 \text{ }\mu\text{m}$  in both figures (the deviations of the actual radii are too small in scale to be seen in figure 7). While the contours at 5 s reveal irregularities in the temperature distribution within the target, the final contours approximately follow the shape of the ice surface, suggesting an isothermal ice surface. This confirms that the system has entered the steady state at the end of the simulation.

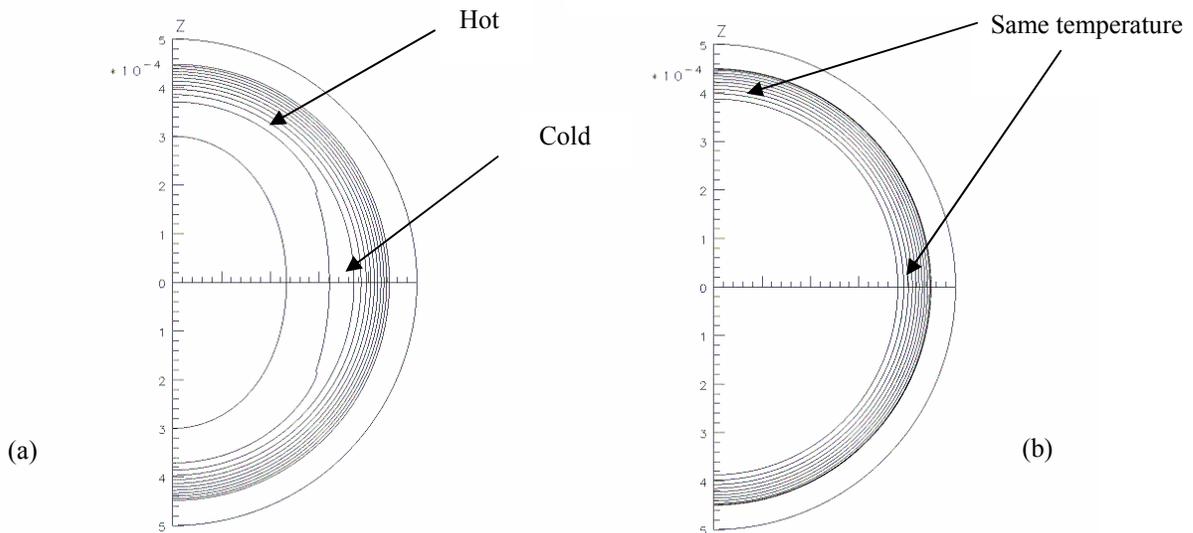


Figure 7: (a) Temperature contours in the target near the beginning of the layering process ( $t=5$  s). The presence of the ring noticeably affects the contours near the equator. The distorted contour shows that the ice surface near the equator is colder than the surface near the poles. (b) Contours after the layering ( $t=1$  h). The z axis is vertical and the outer radius is  $500 \text{ }\mu\text{m}$ . The outer radius of the target itself is  $450 \text{ }\mu\text{m}$ .

Figure 8 illustrates the temperature distribution as a function of radius at the equator ( $\theta = 90^\circ$ ) and at the pole ( $\theta = 0^\circ$ ) in the vicinity of the ice layer, (a) after five seconds and (b) after an hour. Near the beginning of the layering process, there is a small but noticeable difference in temperature between the polar and equatorial regions. After layering, the temperature at the  $D_2$  ice/gas interface is uniform. Also, note the parabolic temperature profile in the ice layer. The temperature gradient increases towards the outer surface because the heat flux at each radius must equal the total IR heat deposited inside that radius. Since the heat flux on either side of the outer ice surface is the same, the low thermal conductivity of the helium gas causes a steep temperature gradient outside the ice. There is also a small temperature gradient in the  $D_2$  gas, because of the infrared heating of the gas. The temperature difference across the ice is about  $500 \mu\text{K}$ , and there

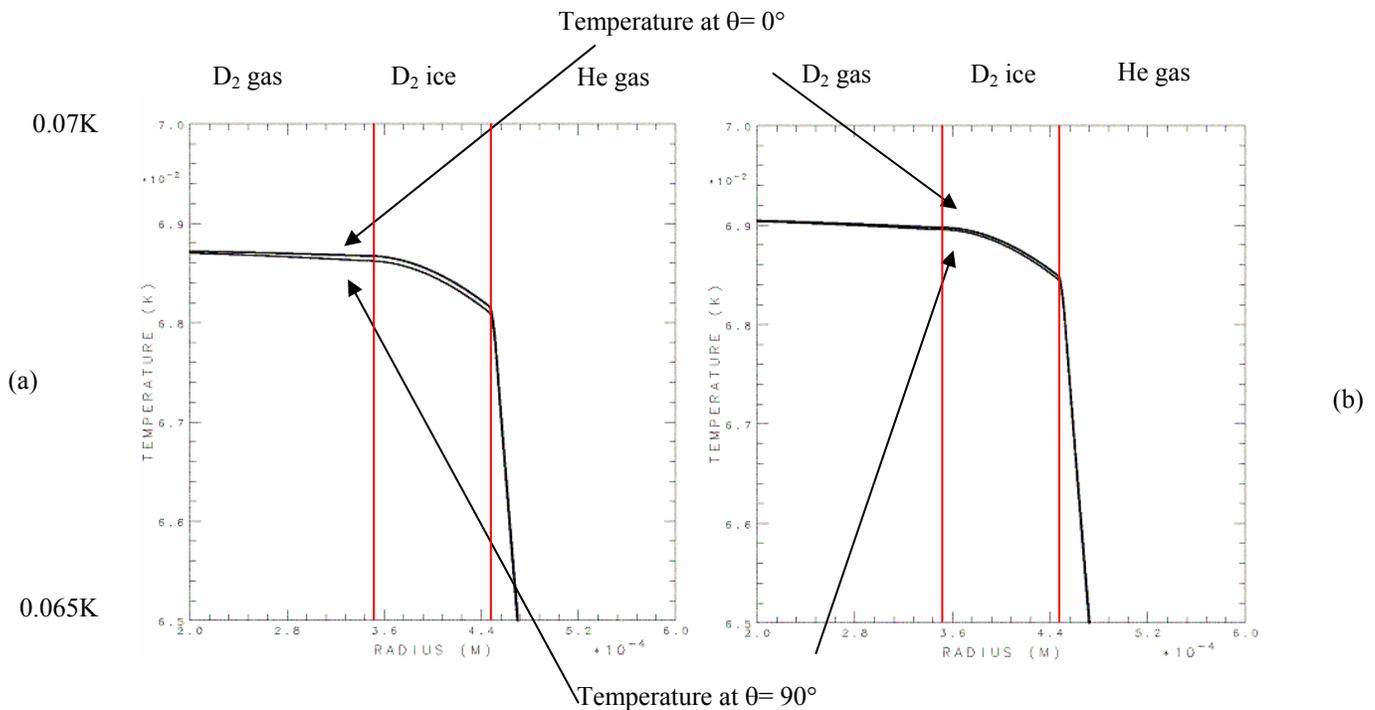


Figure 8: (a) Temperature profile as a function of radius from  $200 \mu\text{m}$  to  $600 \mu\text{m}$  near the beginning of the layering process ( $t=5$  s). (b) Temperature profile after layering ( $t=1$  h). The temperatures on the scale represent the temperature above the reference temperature of  $18.6$  K. The vertical red lines indicate the ice layer (from  $347 \mu\text{m}$  to  $447 \mu\text{m}$ ).

is about 0.1 K of difference between the outside of the target and the edge of the layering sphere. The figure indicates that the ice surface temperature approaches uniformity with respect to  $\theta$  during the layering process.

Figure 9 plots the ice thickness against  $\theta$ . The red line represents the initial thickness, which is uniform, and the blue line indicates the result after layering. The ice is slightly thicker (the interface radius is smaller) in the equatorial region ( $\theta=90^\circ$ ).

Figure 9 shows that the maximum interface radius after layering is 347.6  $\mu\text{m}$ , which occurs at the poles ( $\theta=0^\circ$  or  $\theta=180^\circ$ ). The minimum radius, reached at the equator ( $\theta=90^\circ$ ), is 346.3  $\mu\text{m}$ . The largest deviation from the mean radius, which happens at the equator, is about 0.7  $\mu\text{m}$ . As the deviation is within the design tolerance of 1  $\mu\text{m}$ , we may conclude that the ring does not pose a significant threat to ice uniformity.

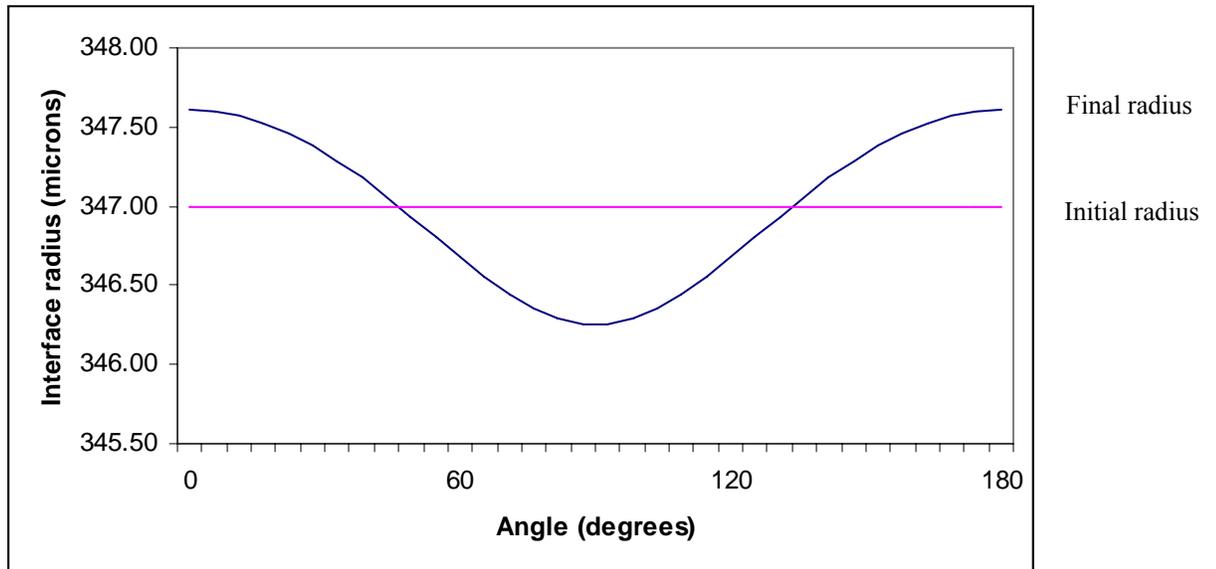


Figure 9: Ice-gas interface radius for different angles  $\theta$ . The red and blue lines represent the initial and the final radii, respectively. The initial radius was set to be uniform as a function of  $\theta$ . After layering, the ice becomes slightly thicker (the interface radius slightly smaller) near the equator ( $\theta = 90^\circ$ ) and slightly thinner at the poles ( $\theta = 0^\circ$  and  $\theta = 180^\circ$ ).

#### 4. Conclusion

A new computer program has been developed to simulate the layering of a cryogenic target. A set of nonlinear partial differential equations was established to model the layering process. The equation set was decomposed into two parts, a linear partial differential equation set and a few linear and nonlinear functions. This significantly reduces computation complexity without noticeably sacrificing calculation accuracy. The interface cells are modeled as a linear mixture of gas and ice.

The program was applied to determine the ice shape after layering for “Saturn ring” targets. Two schemes were developed to reduce the number of time steps. First, the problem was cast into the framework of finding the steady state solution and the calculation was expedited by starting from an appropriate initial condition. Second, a variable time step size was implemented. A larger step size was adopted when the system is slowly varying to ensure computational efficiency, and a smaller time step is used when the system is rapidly changing to avoid large approximation errors.

The result showed that the nonuniformities caused by the plastic rings are small. The largest deviation from the mean radius is about  $0.7 \mu\text{m}$ . This deviation is within the design tolerance of  $1 \mu\text{m}$ , so its impact on fusion performance is acceptable.

#### 5. Acknowledgements

I would especially like to thank Dr. Craxton for giving me the opportunity to participate in this wonderful program, as well as for the endless hours spent helping me with my project. I would also like to thank the fellow students in the LLE High School Summer Research Program for making this summer such a memorable experience.

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**Appendix. List of Physical Constants Used<sup>6,7</sup>**

	$K$ (W/mK)	$C_v$ (J/(m <sup>3</sup> K))	$Q_{IR}$ (W/m <sup>3</sup> )	$L$ (J/molecule)
Solid D <sub>2</sub>	0.38	$5.802 \times 10^5$	$4.6 \times 10^4$	$9.80 \times 10^{-22}$
Plastic	0.05	$1.95 \times 10^5$		
Helium	0.0227	0.969		
D <sub>2</sub> gas	0.009	2155	4.6	

$T_{ref}$	18.6 K
$N_0$	$4.89 \times 10^{22}$ molecules/(m <sup>2</sup> s)
$\alpha$	$2.43 \times 10^{22}$ molecules/(m <sup>2</sup> sK)
$\beta$	$7.84 \times 10^{-4}$ m/s
$\gamma$	$-2.11 \times 10^{-5}$ m/(sK)