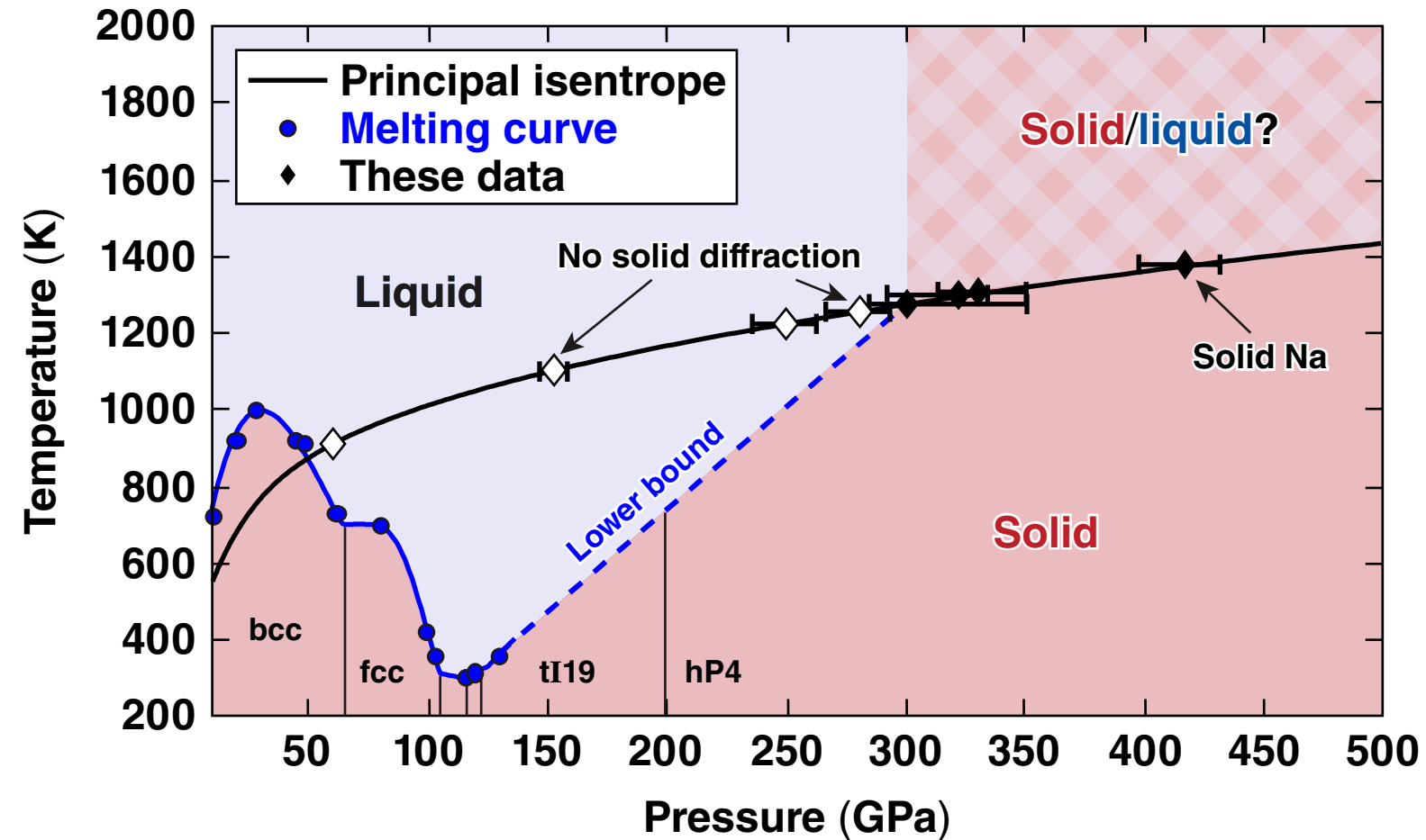


# X-Ray Diffraction of Ramp-Compressed Sodium



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## Summary

# Melting and resolidification of Na to a solid phase is observed for pressures greater than 300 GPa



- Alkali metals, including Na, serve as a prototypical system to guide theory for ultradense matter beyond the Thomas–Fermi model
- High-power lasers ramp compressed Na and nanosecond *in-situ* x-ray diffraction (XRD) was used to measure the crystal structure at pressures up to 418 GPa
- Solid Na was observed at 301 to 418 GPa; therefore, the melting temperature increases dramatically above 140 GPa
- Simultaneous XRD and optical reflectivity measurements reveal that Na remains reflective to at least 290 GPa, in striking contrast to room-temperature static-compression measurements

# Collaborators

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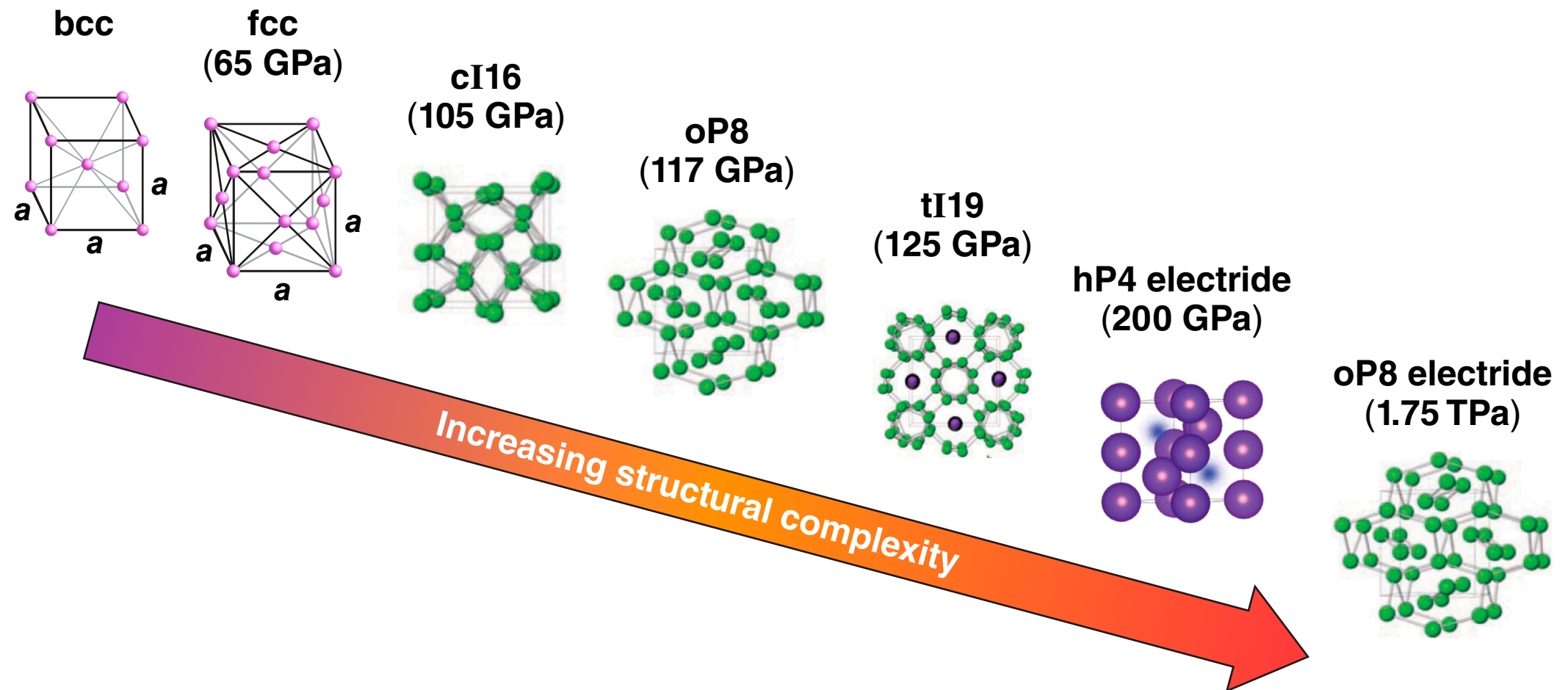
**M. I. McMahon**  
**University of Edinburgh**

# Motivation

Alkali metals, including sodium, are an ideal test bed for studying high-pressure structural phenomenon



1	-1, +1	<b>H</b>
Hydrogen 1.008		
3	+1	<b>Li</b>
Lithium 6.941		
11	+1	<b>Na</b>
Sodium 22.990		
19	+1	<b>K</b>
Potassium 39.098		
37	+1	<b>Rb</b>
Rubidium 85.468		
55	+1	<b>Cs</b>
Cesium 132.905		
87	1	<b>Fr</b>
Francium 223.020		

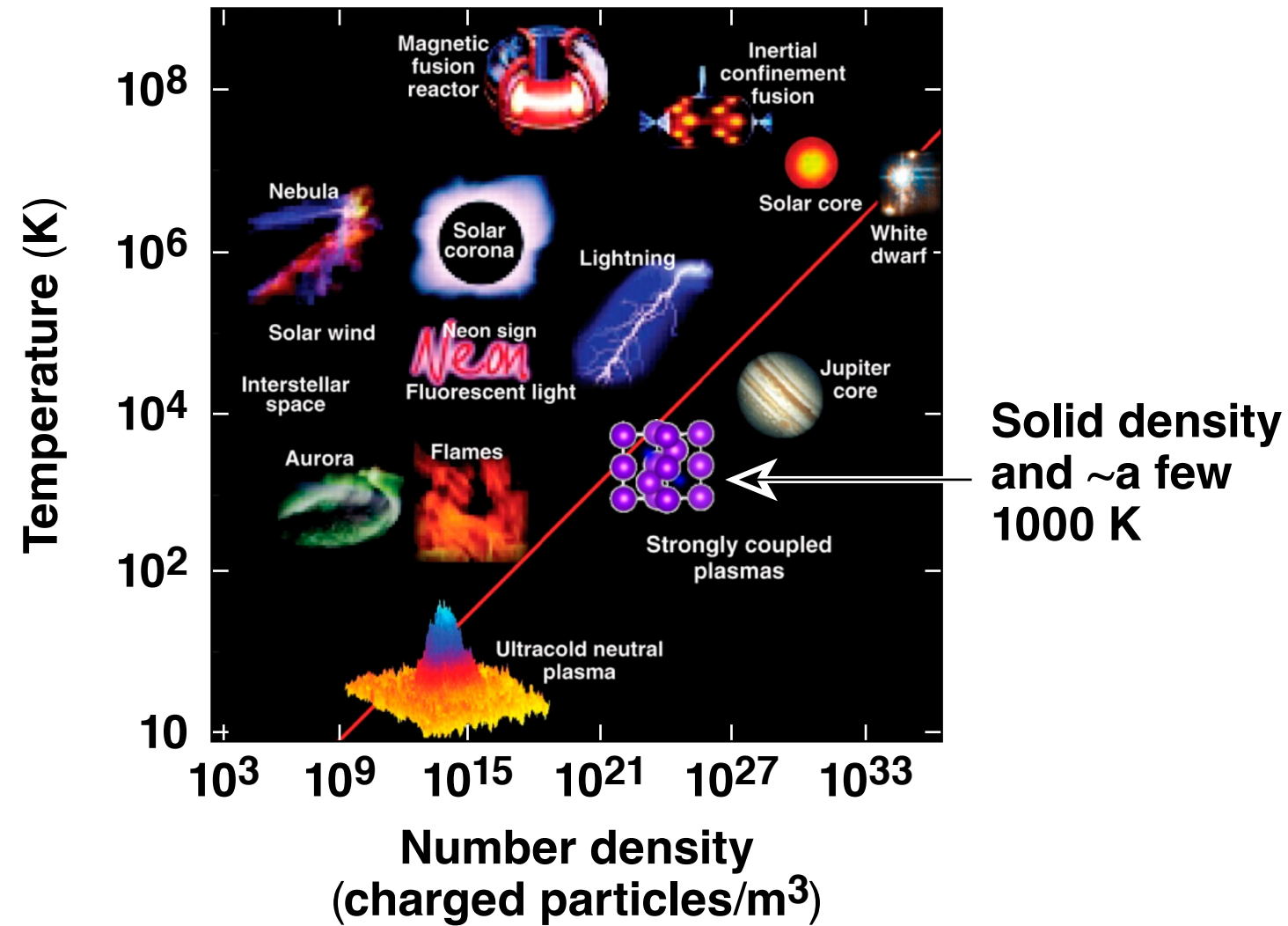


bcc: body centered cubic  
fcc: face centered cubic

Y. Ma *et al.*, Nature **458**, 182 (2009);  
E. Gregoryanz *et al.*, Science **320**, 1054 (2008).

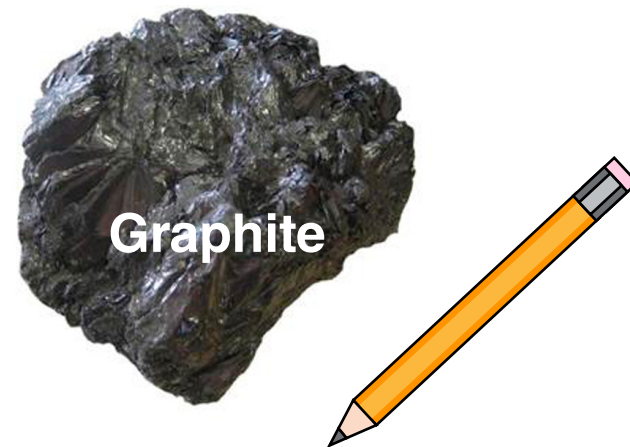
E26646g

# High-energy-densities are achieved in these experiments through ramp-compression



# Atomic structure affects many properties of materials

## (Carbon) crystals

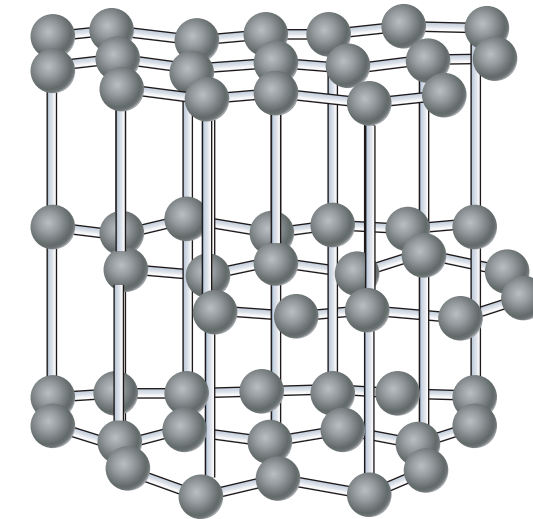


## Physical properties

Density  
 $2.2 \text{ g/cm}^3$

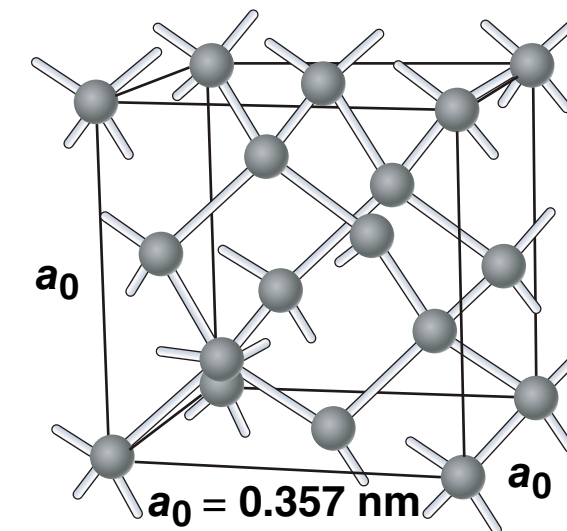
Yield strength  
 $<0.2 \text{ GPa}$

## Atomic structure



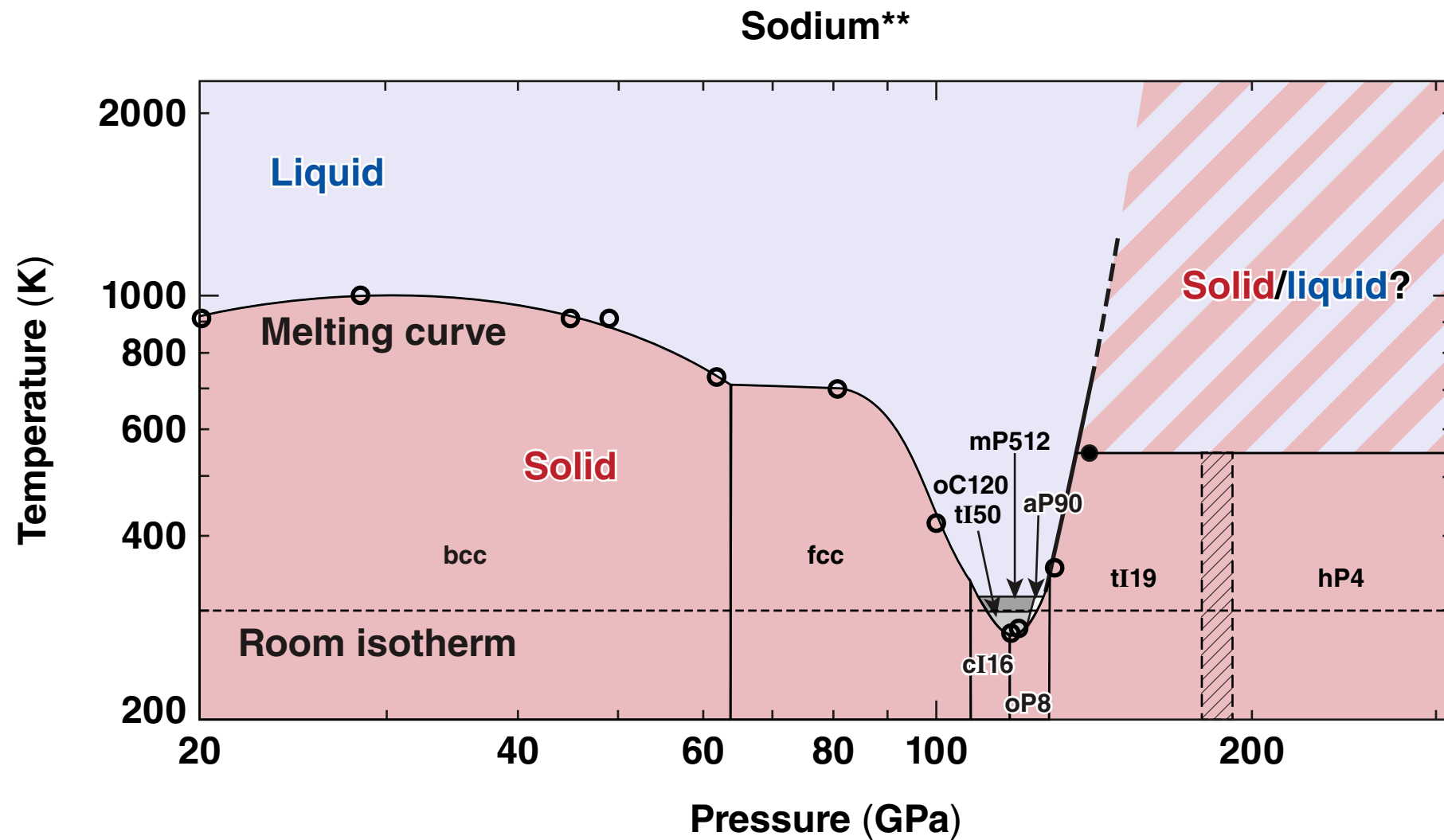
Density  
 $3.51 \text{ g/cm}^3$

Yield strength  
 $>110 \text{ GPa}$



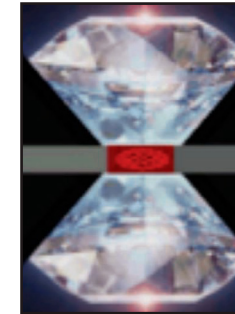
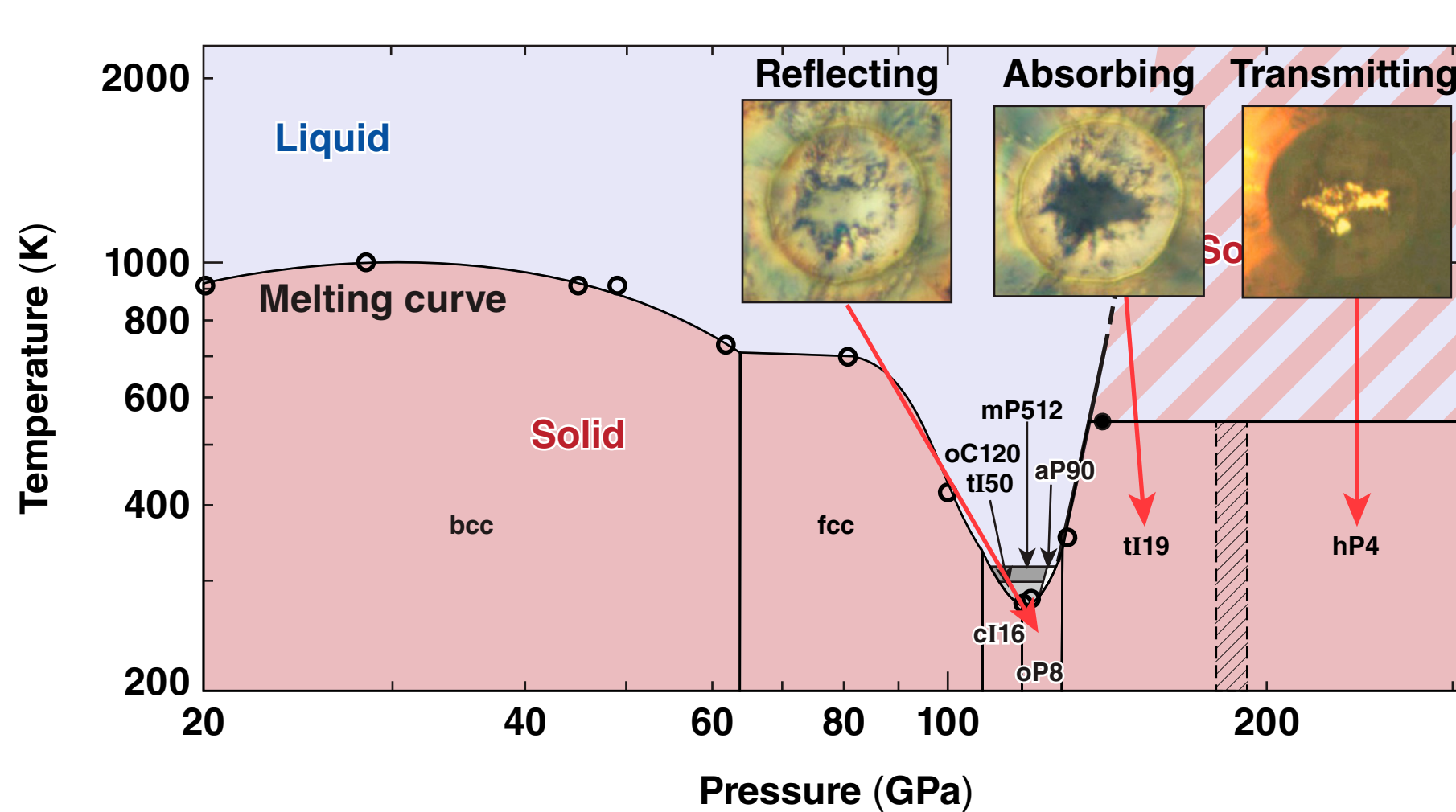
There is a 1.6× difference in density, 10× difference in compressibility, and 500× difference in strength.

# At high pressures, Na has a unique melting curve that exhibits a minimum at 120 GPa, then rises steeply



\*E. Gregoryanz *et al.*, *Science* **320**, 1054 (2008).  
\*\*M. Marqués *et al.*, *Phys. Rev. B* **83**, 184106 (2011).

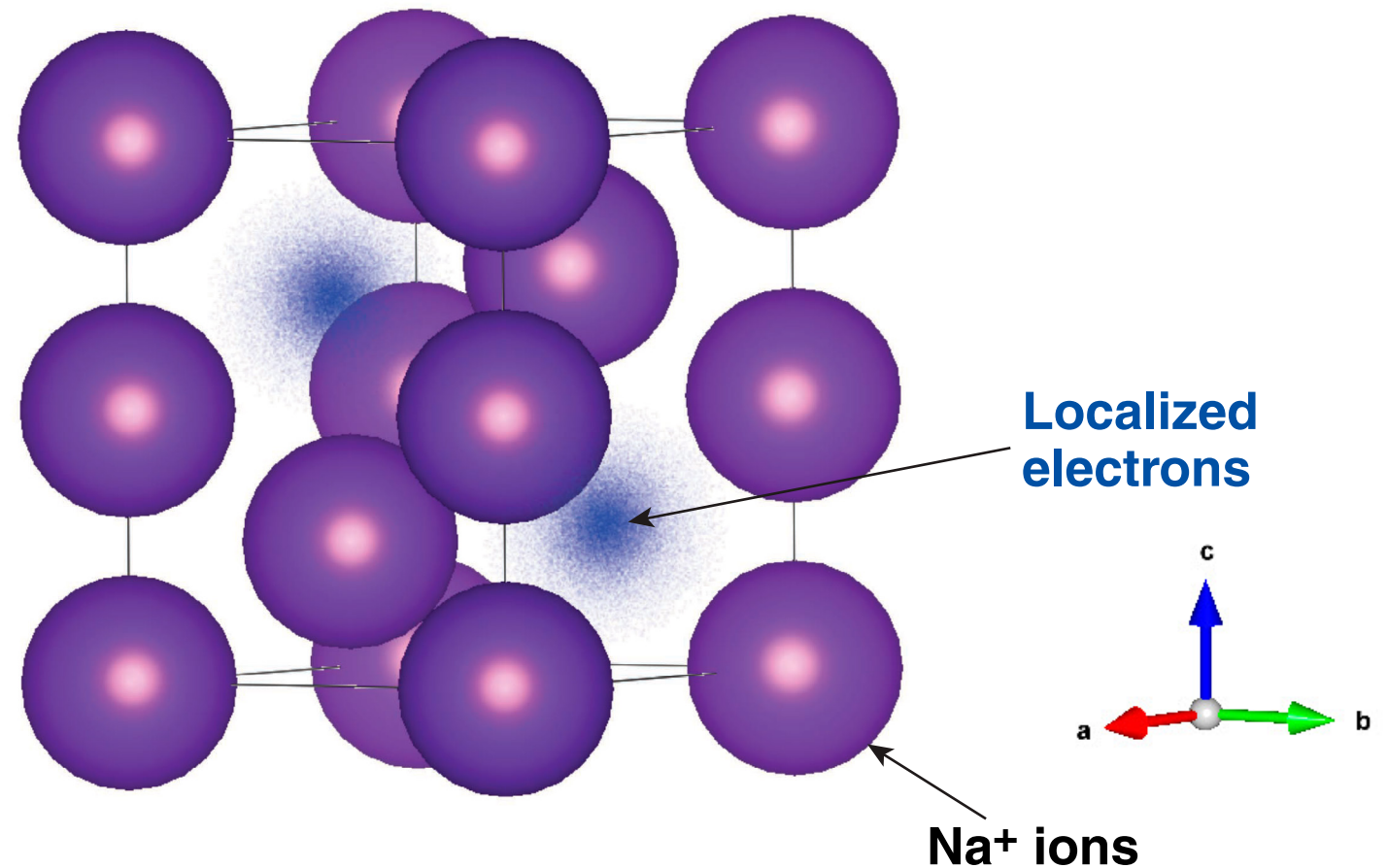
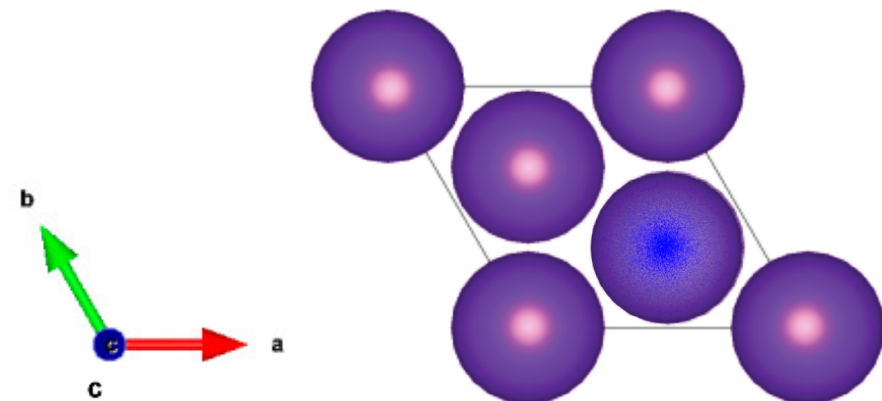
# Diamond-anvil cell (DAC) experiments\* show that Na transforms into an optically transparent phase at 200 GPa



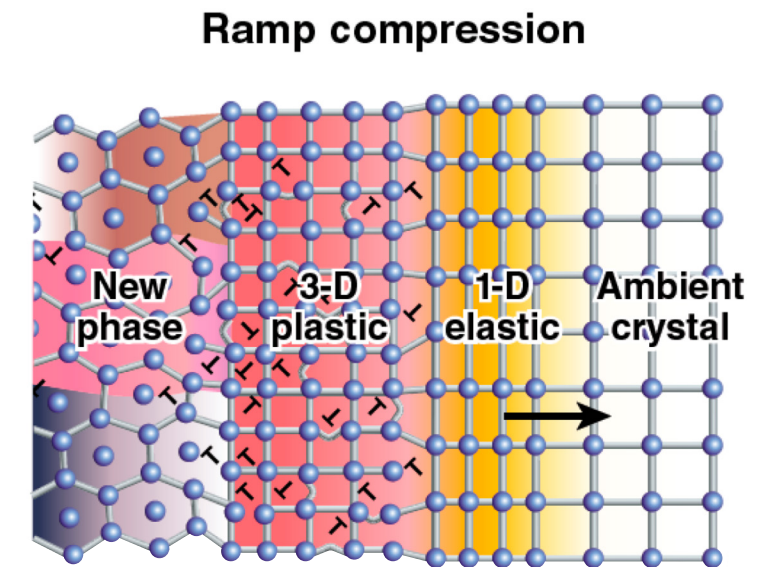
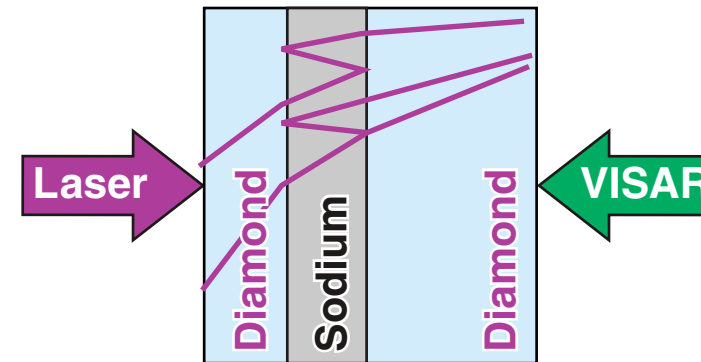
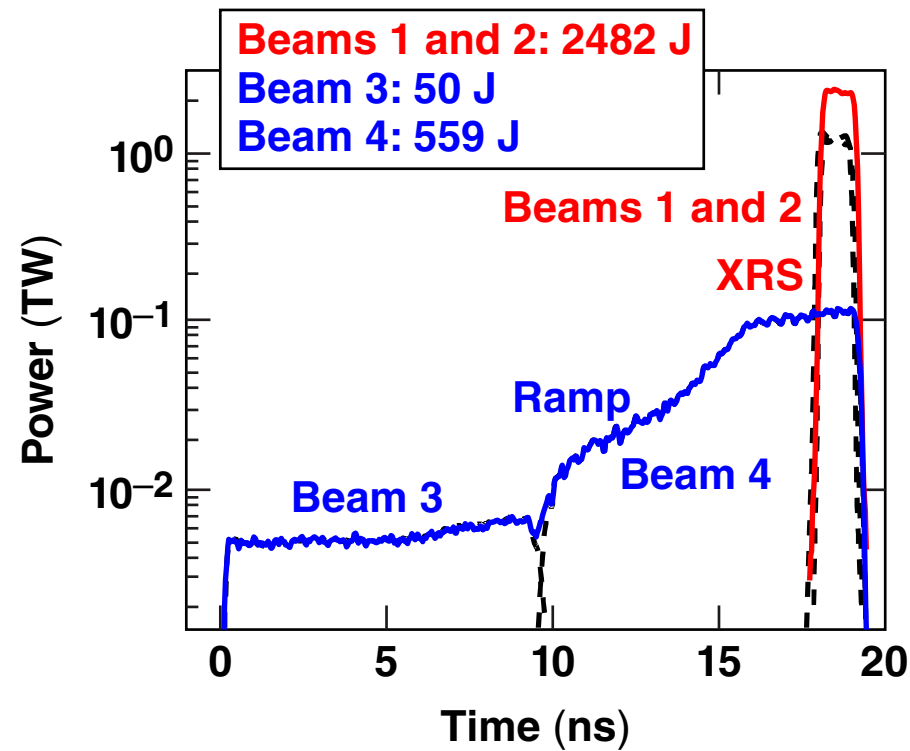


The transparent phase is predicted\* to be an electride hP4 structure, where conduction electrons are “trapped” in interstitial wells, producing an insulating behavior

hP4: double hexagonal close-packed (dhcp) structure, squeezed more than twice along the c axis

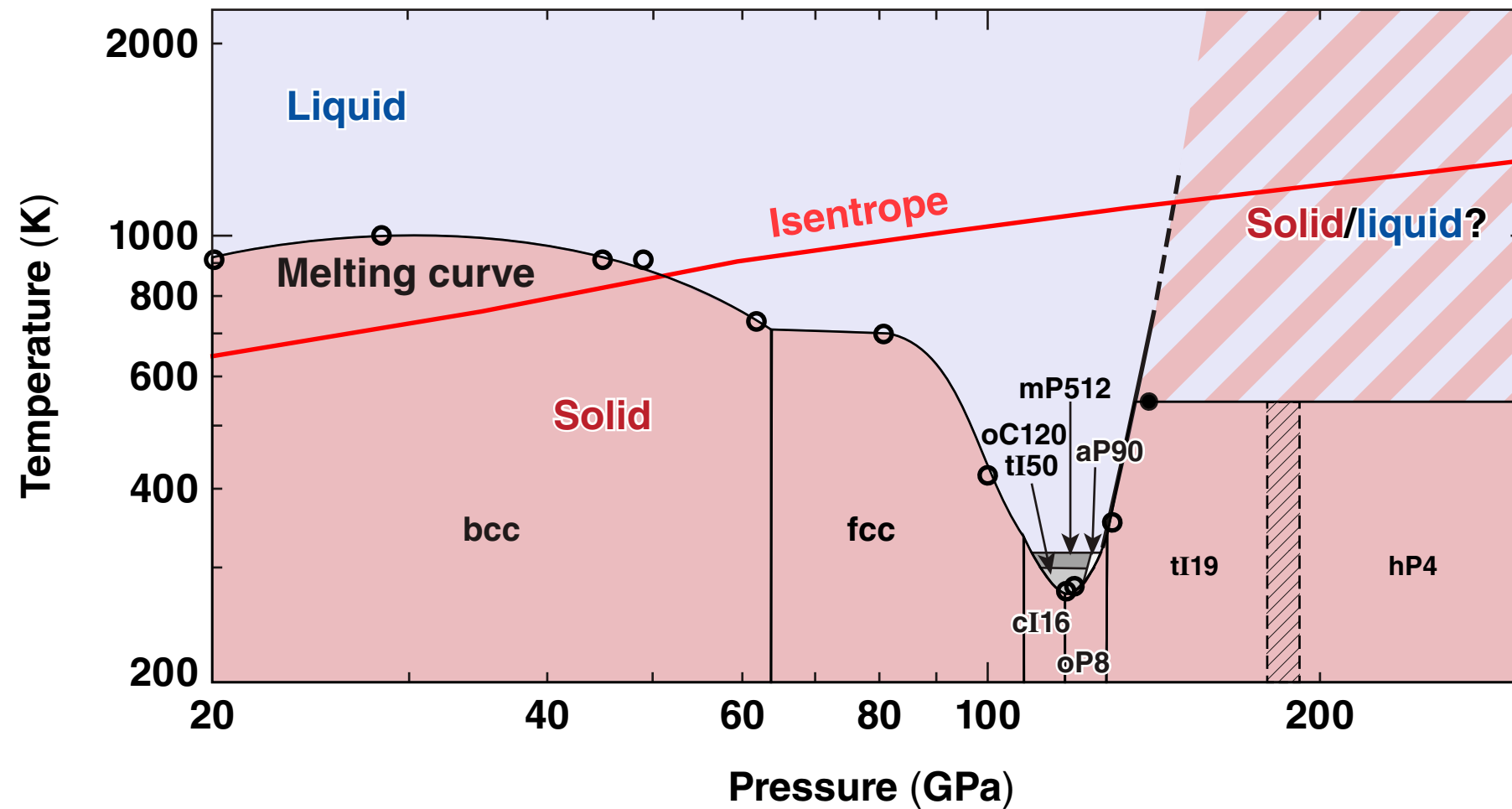


# Ramp compression is used to follow a nearly isentropic compression path to investigate the solid phases of Na

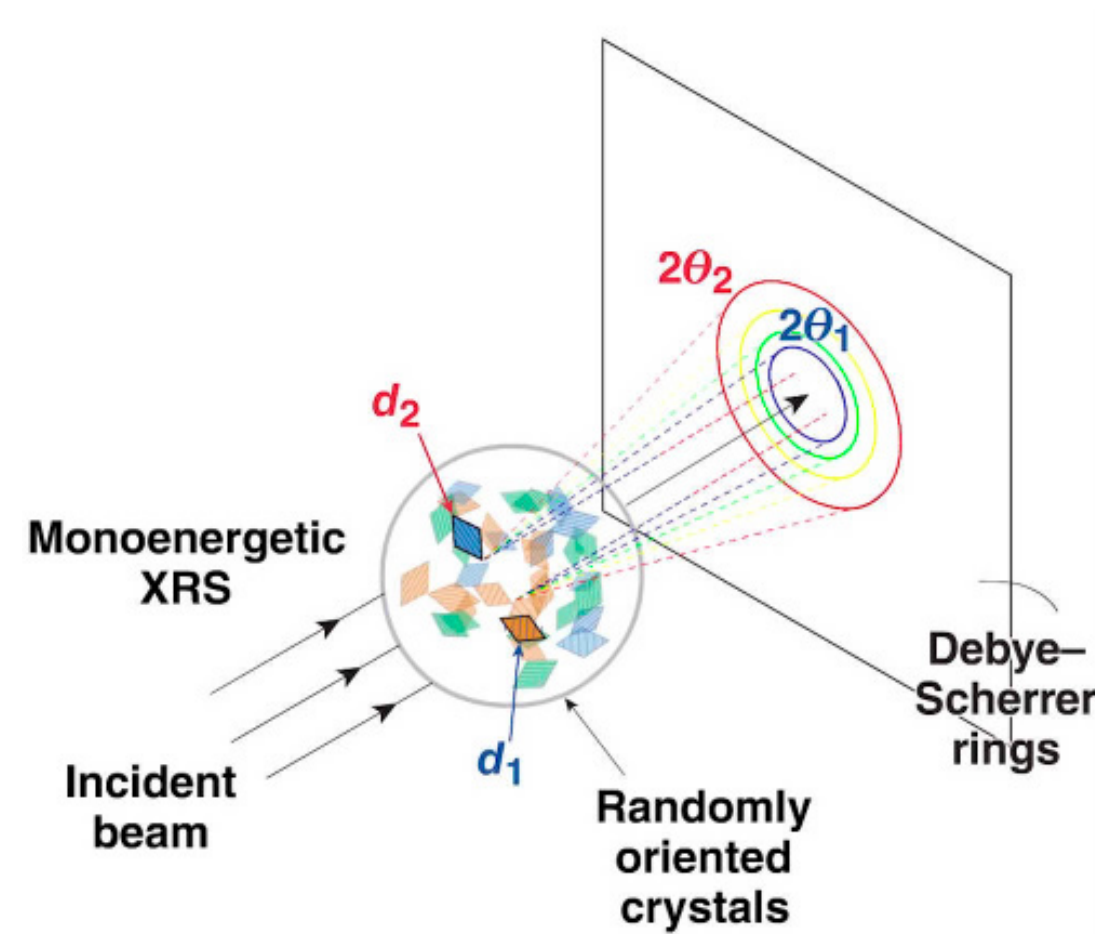
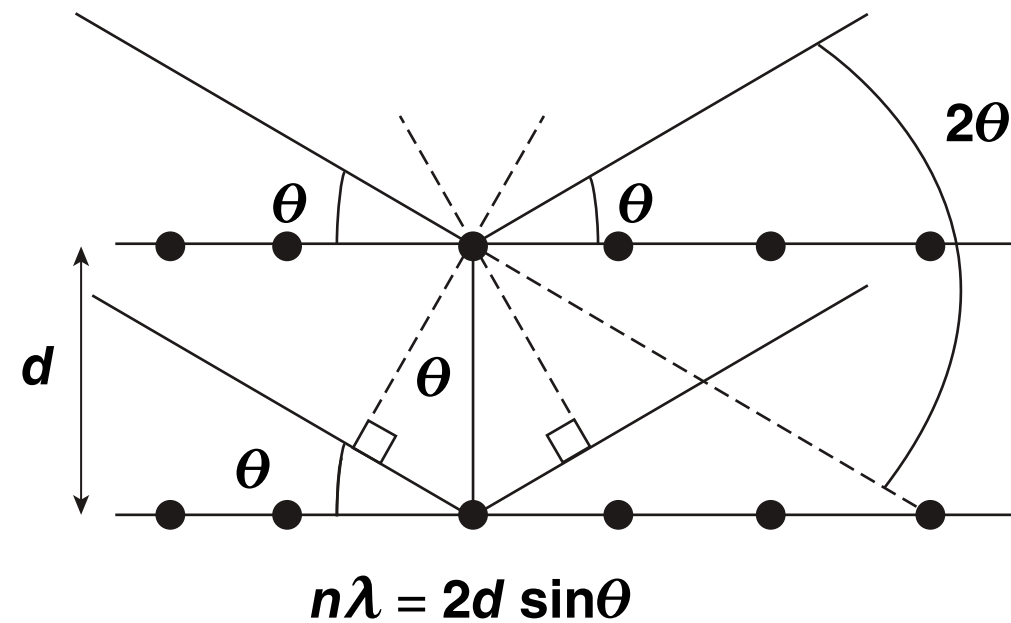


VISAR: velocity interferometer system for any reflector  
 XRS: x-ray source

Assuming isentropic compression, Na will melt in the bcc phase and a solid phase will be observed if the melting curve continues to sharply increase

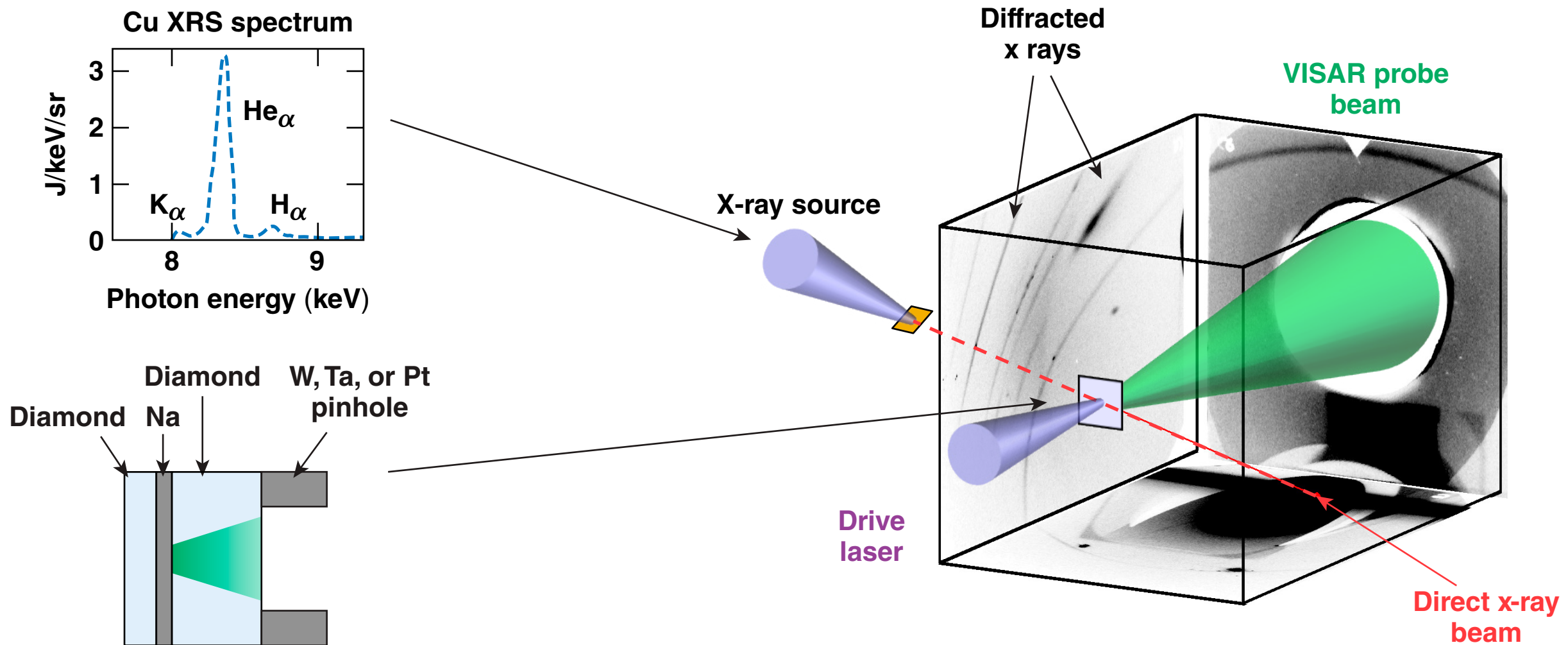


# Monoenergetic x rays incident on an ideal powder sample will diffract in rings of uniform intensity at angles $2\theta$ with respect to the x-ray beam

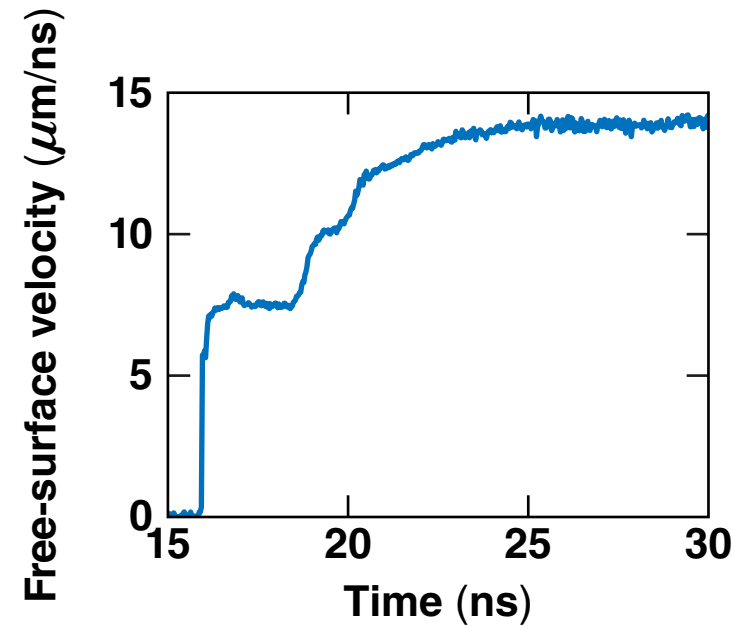
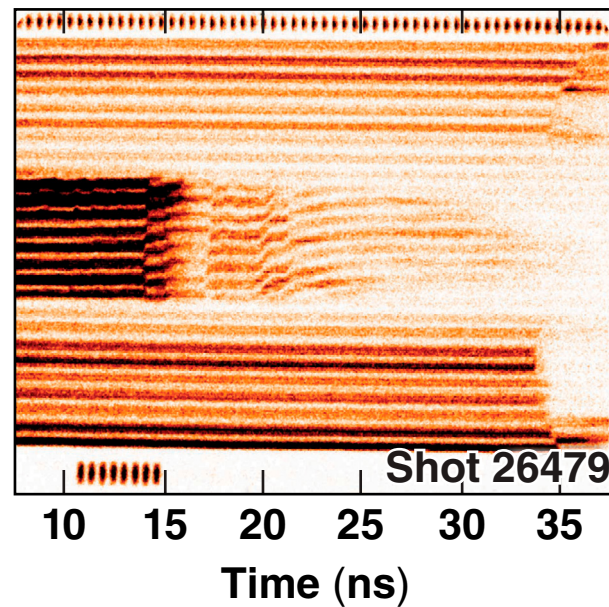
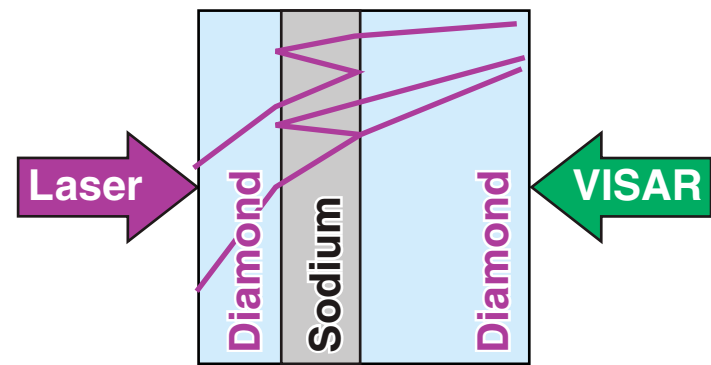


Compression of interatomic spacing ( $d$  spacing) is determined by measuring diffraction angles ( $2\theta$ ) and x-ray wavelength ( $\lambda$ ).

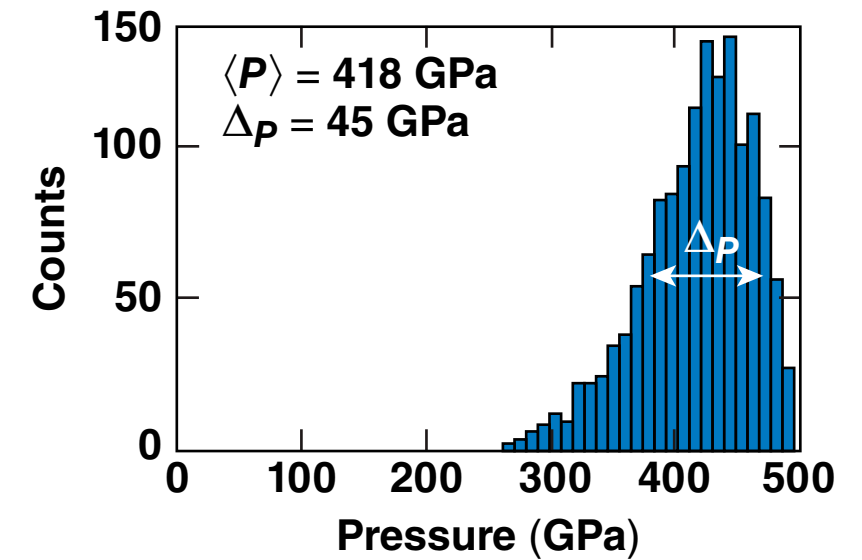
# The powder x-ray diffraction image-plate (PXRDIP\*) platform is used to record diffraction patterns on OMEGA EP



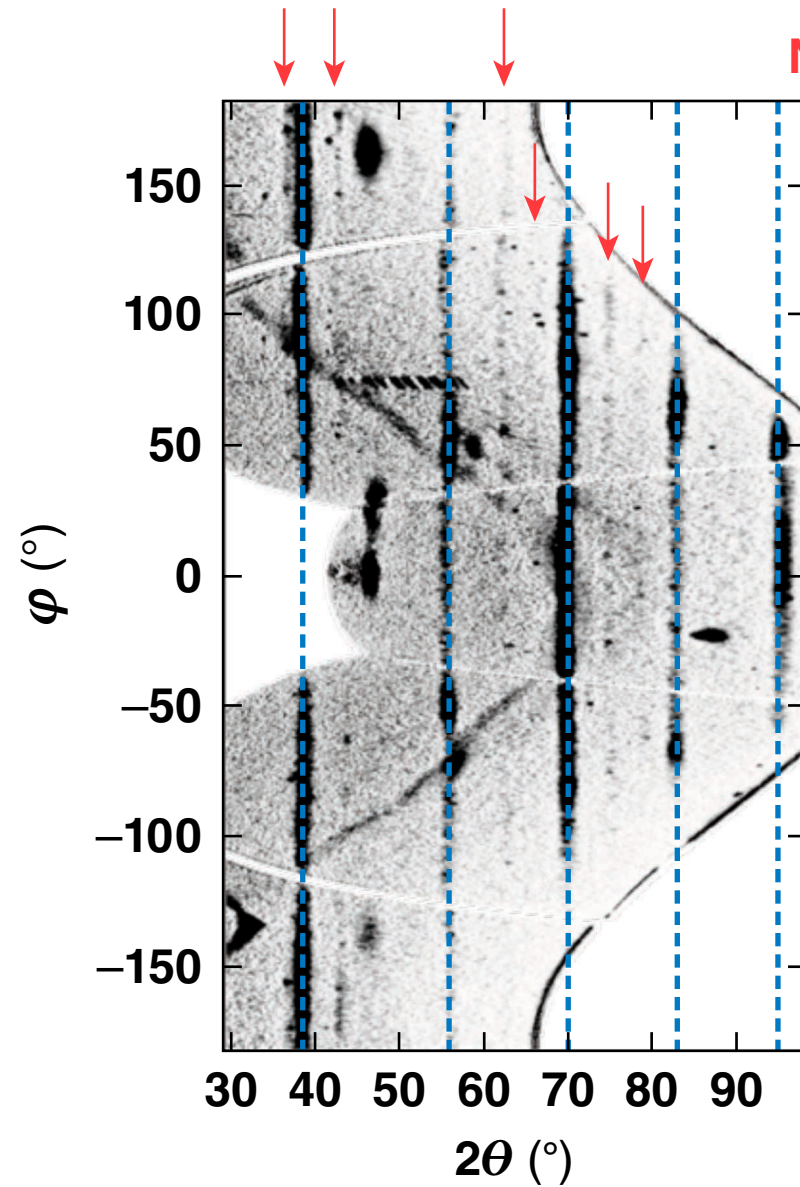
# The velocity of the diamond free surface is back-propagated to determine the pressure in the Na sample



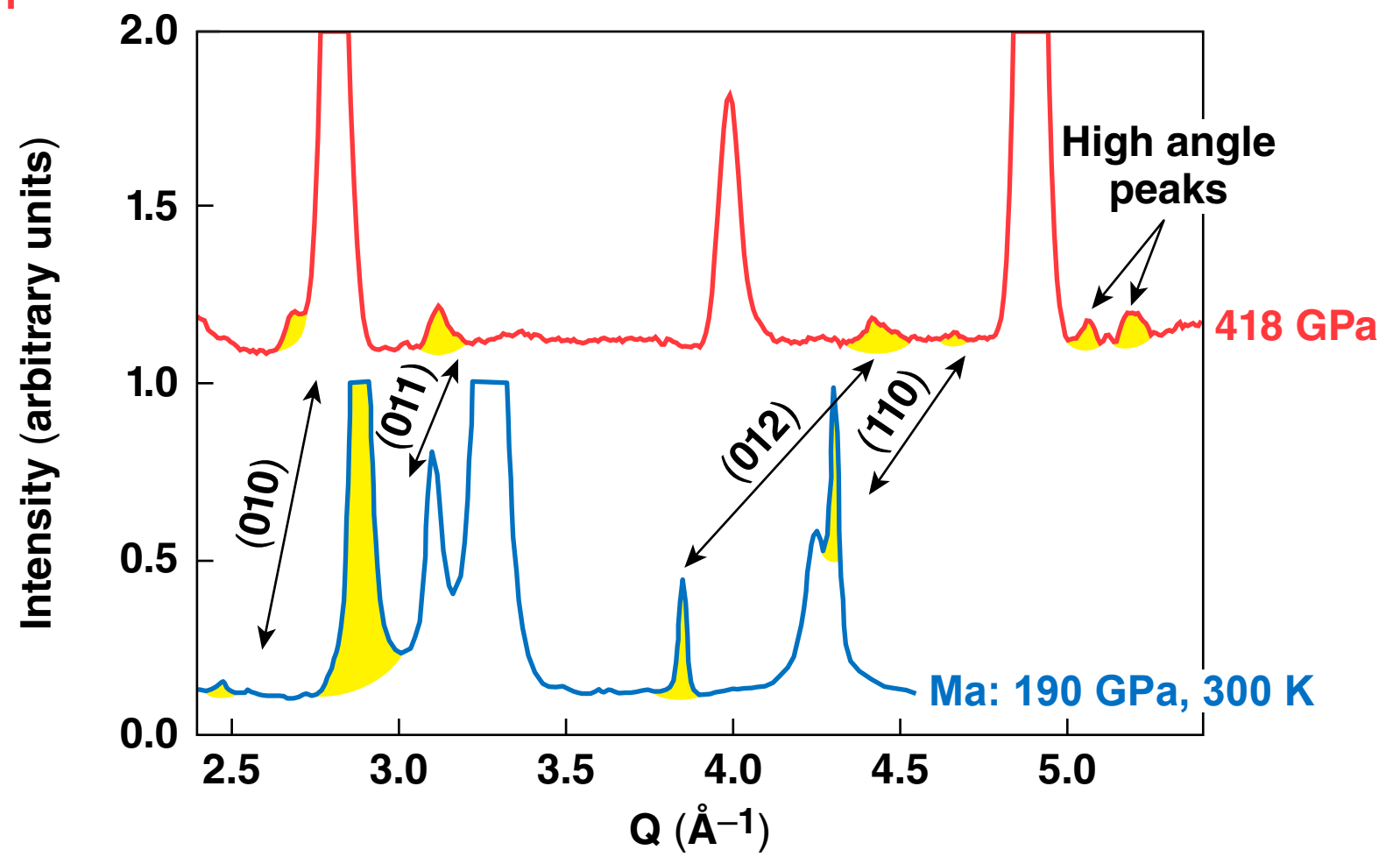
Backward  
characteristics  
algorithm



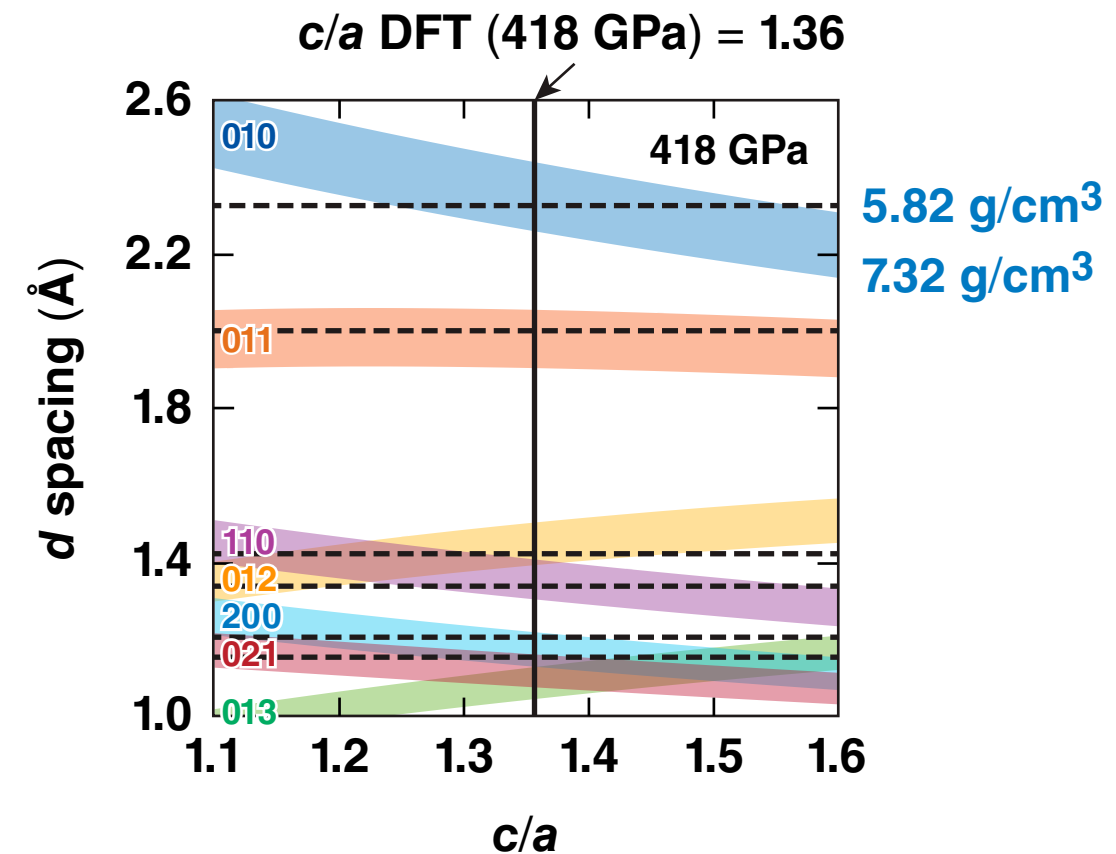
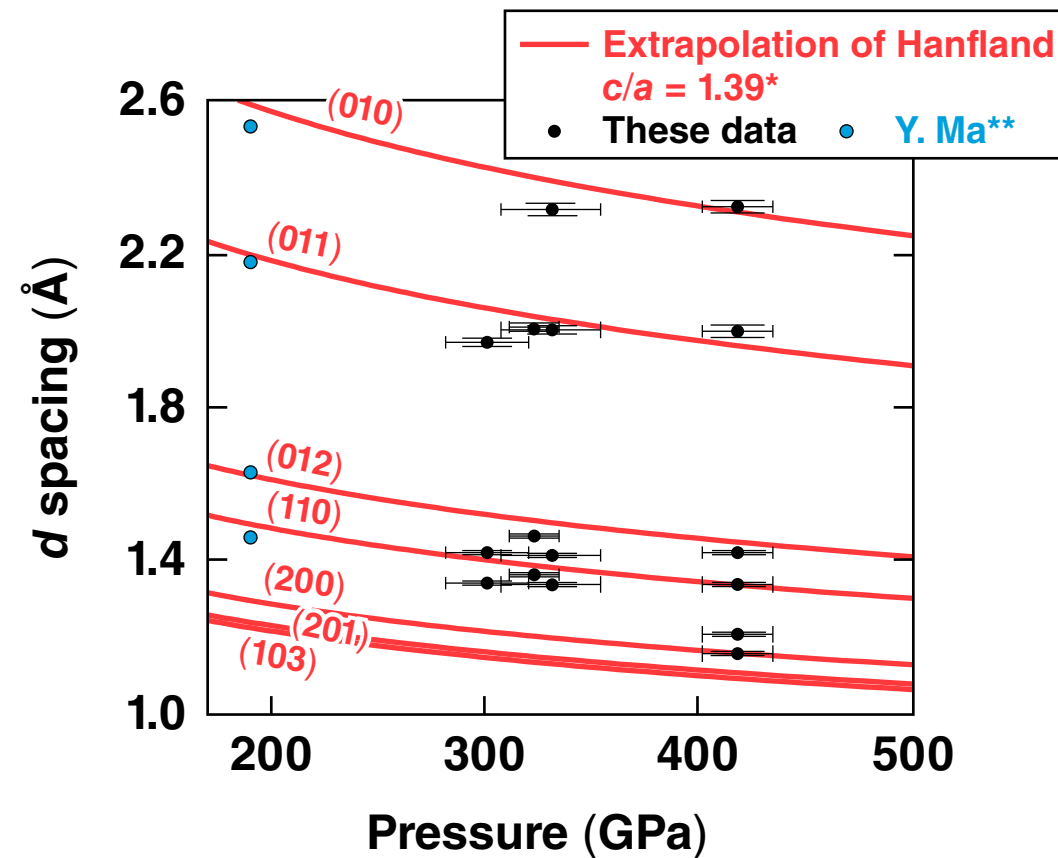
# Evidence of solid Na was observed at 418 GPa with up to six diffraction peaks from compressed Na



Na diffraction lines



# A single, reproducible high-pressure solid phase is observed in Na ramp compressed to 300 to 418 GPa



**This phase exhibits sixfold compression and is not fully consistent with the hP4 phase.**

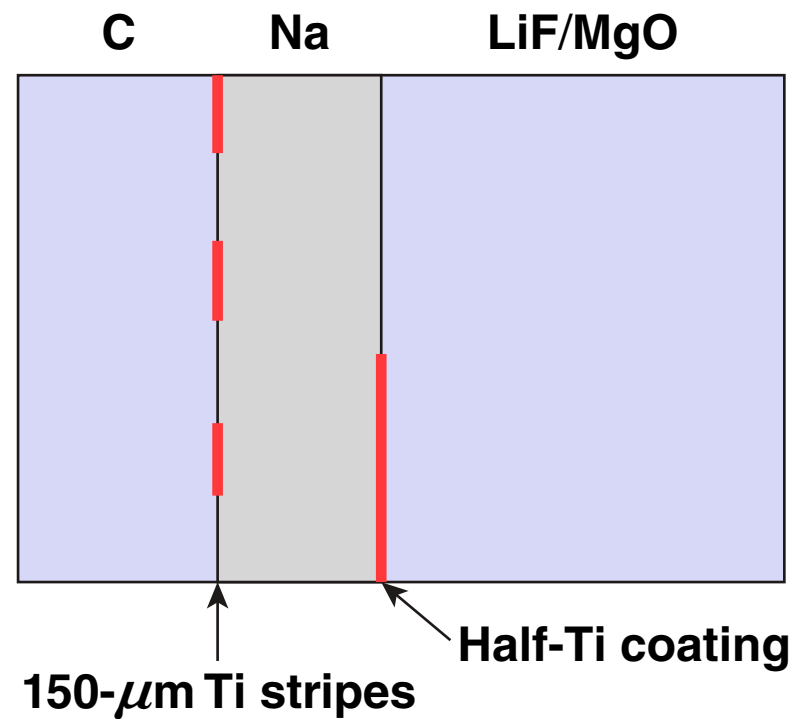
DFT: density functional theory

\*M. Hanfland *et al.*, Phys. Rev. B. **65**, 184109 (2002).

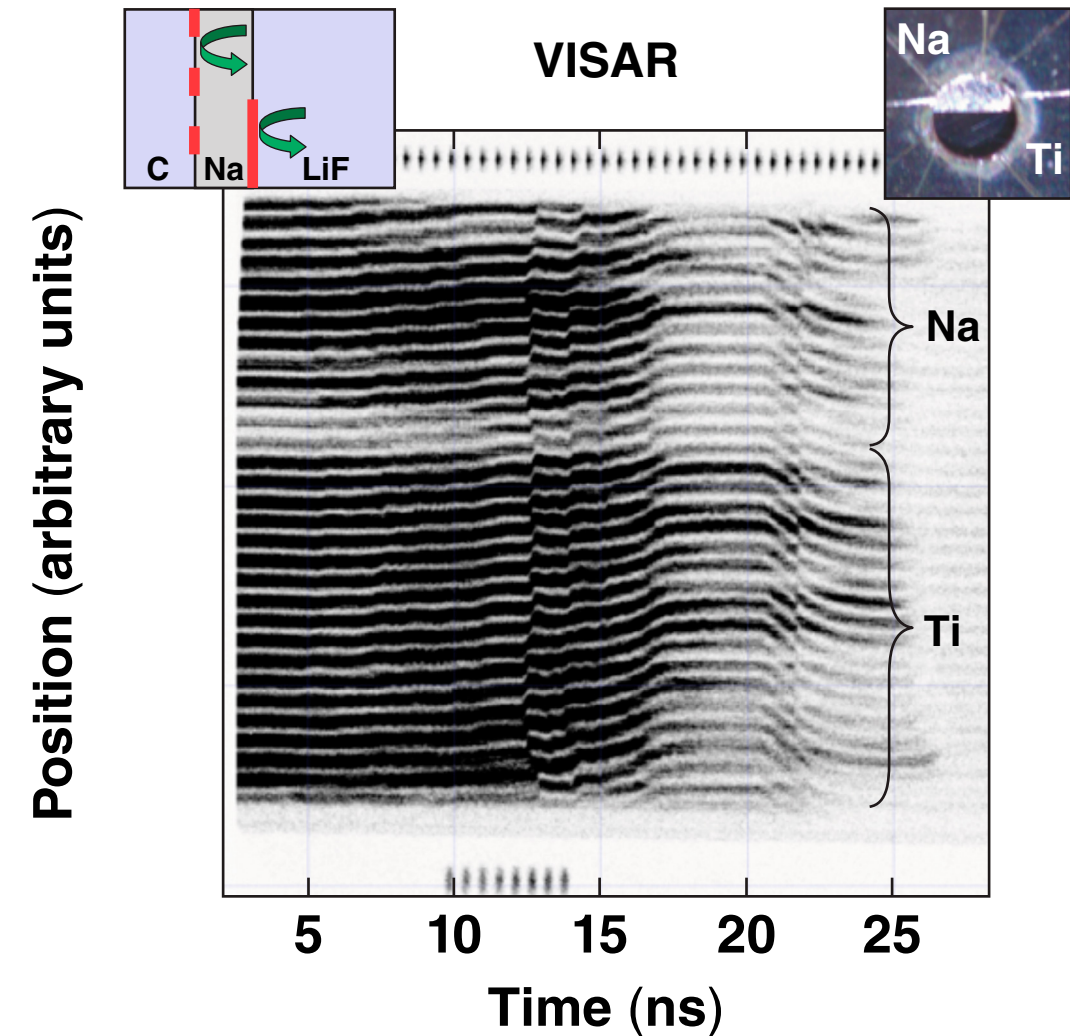
\*\*Y. Ma *et al.*, Nature **458**, 182 (2009)



# A transparent window with a reflectance pattern of Ti stripes is used to detect Na transparency at 532 nm; Na remains reflective to at least 290 GPa



Elevated temperatures may cause delocalization of the electrons and thermal population of the conduction band in Na.



# A simple semiconducting Drude picture is used to constrain the bandgap and temperature of dense Na

Temperature-activated carrier concentration

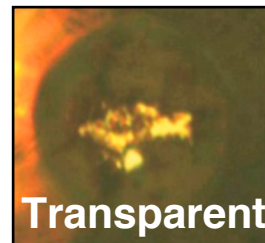
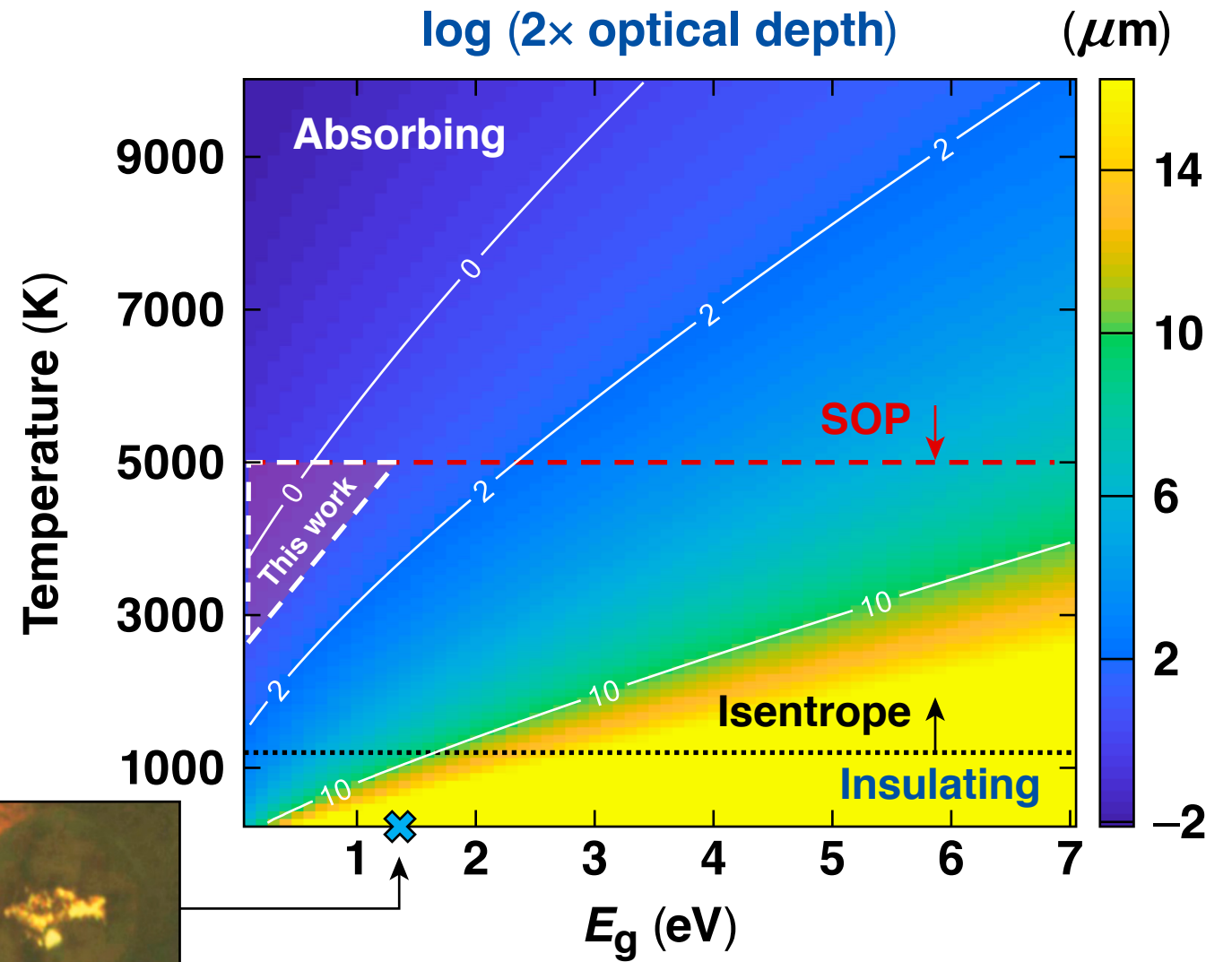
$$N_e = 2 \left( \frac{m_{\text{eff}} k_B T}{2\pi\hbar^2} \right)^{\frac{3}{2}} F(-E_g/2k_B T)$$

Dielectric function

$$\epsilon = \epsilon_b + \frac{i}{\omega\epsilon_0} \sigma$$

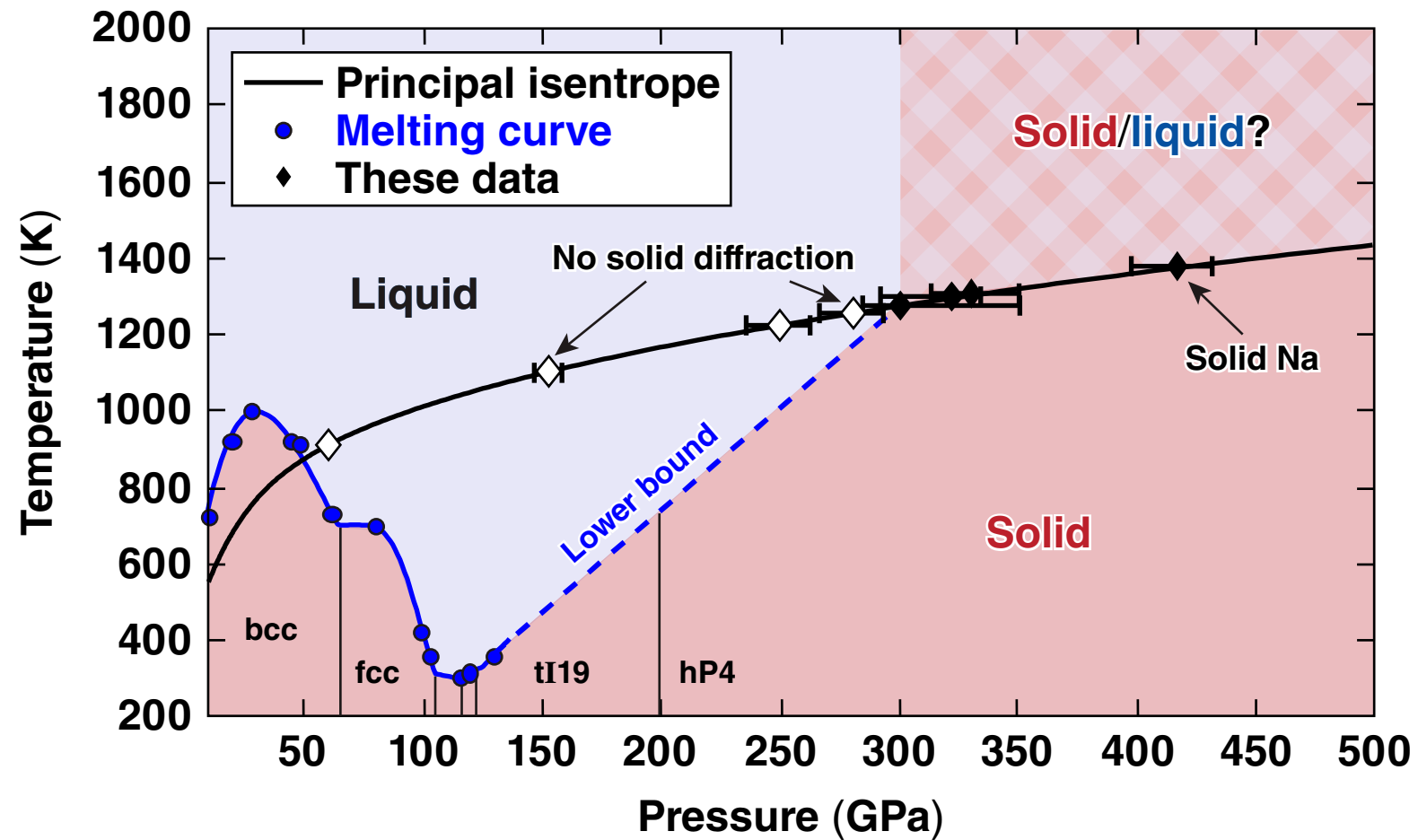
Bound electrons

Free carriers  
(Drude conductivity)



Y. Ma *et al.*, Nature **458**, 182 (2009).  
SOP: streaked optical pyrometer

# We are mapping out the Na phase diagram to 500 GPa

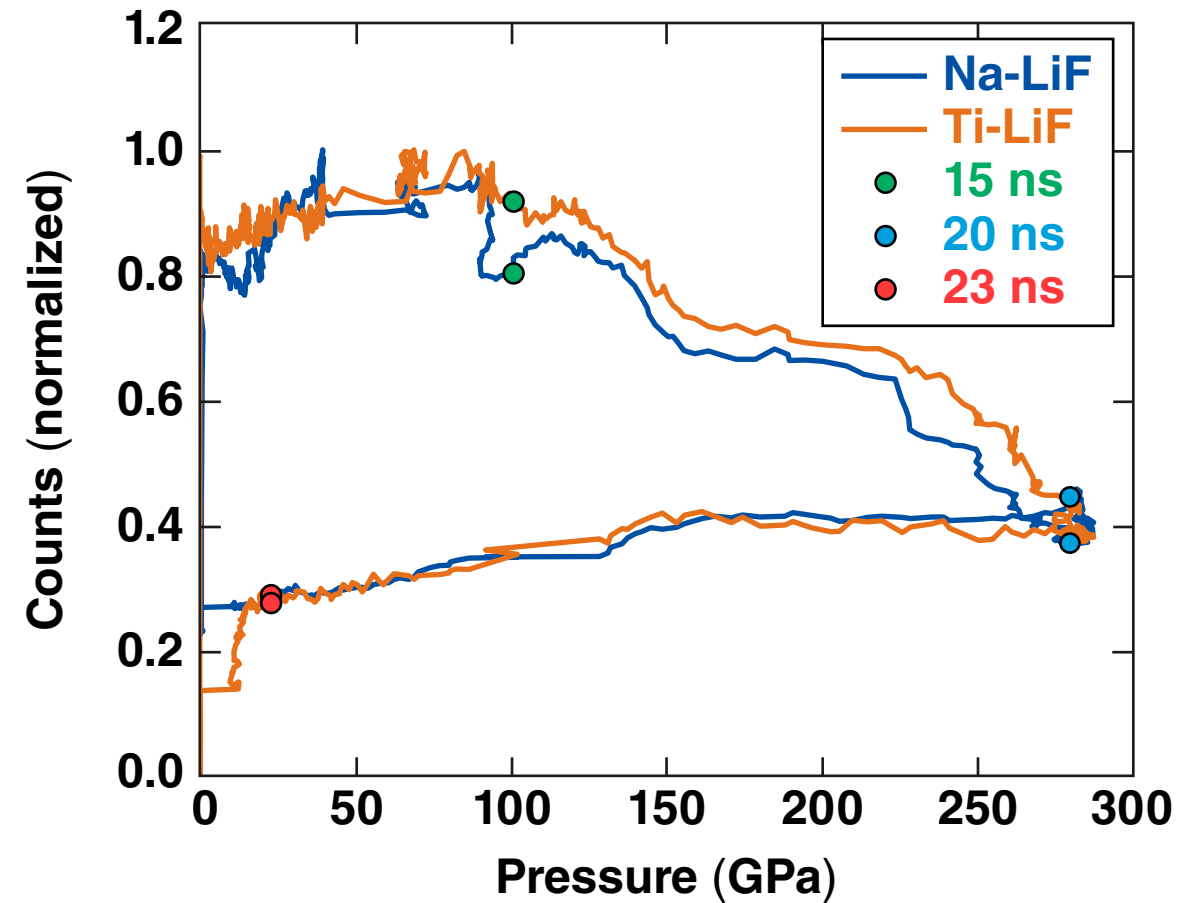
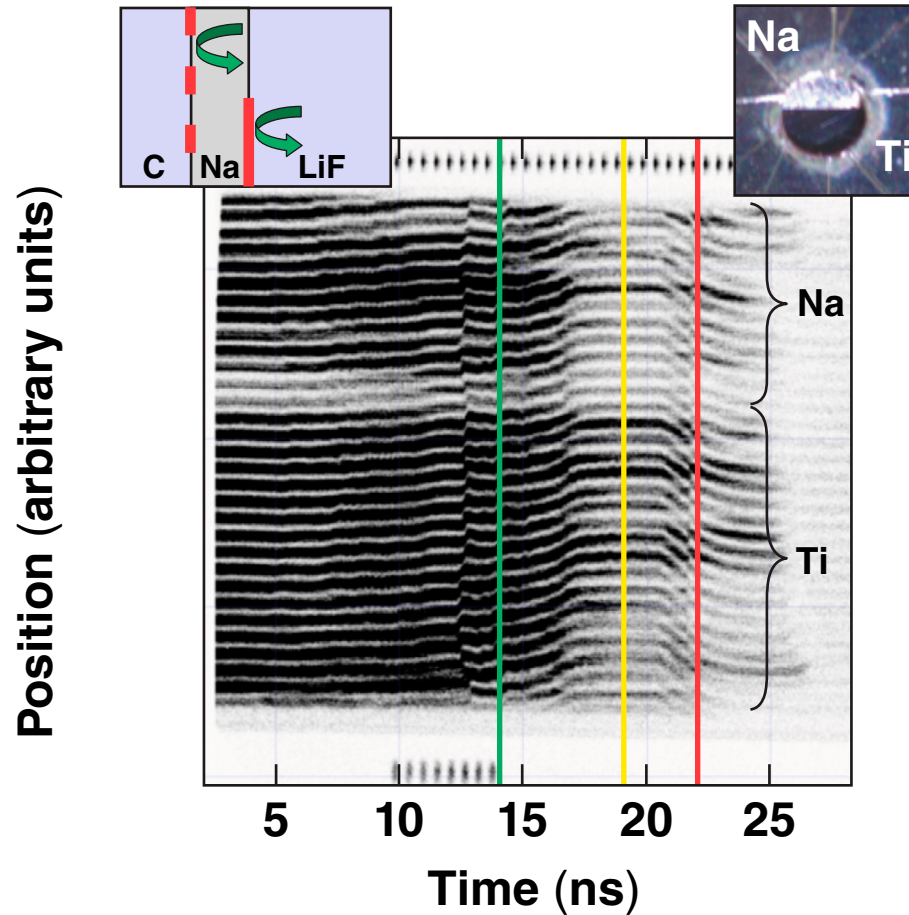


A dramatic increase in the melting temperature occurs above 140 GPa.

# Melting and resolidification of Na to a solid phase is observed for pressures greater than 300 GPa

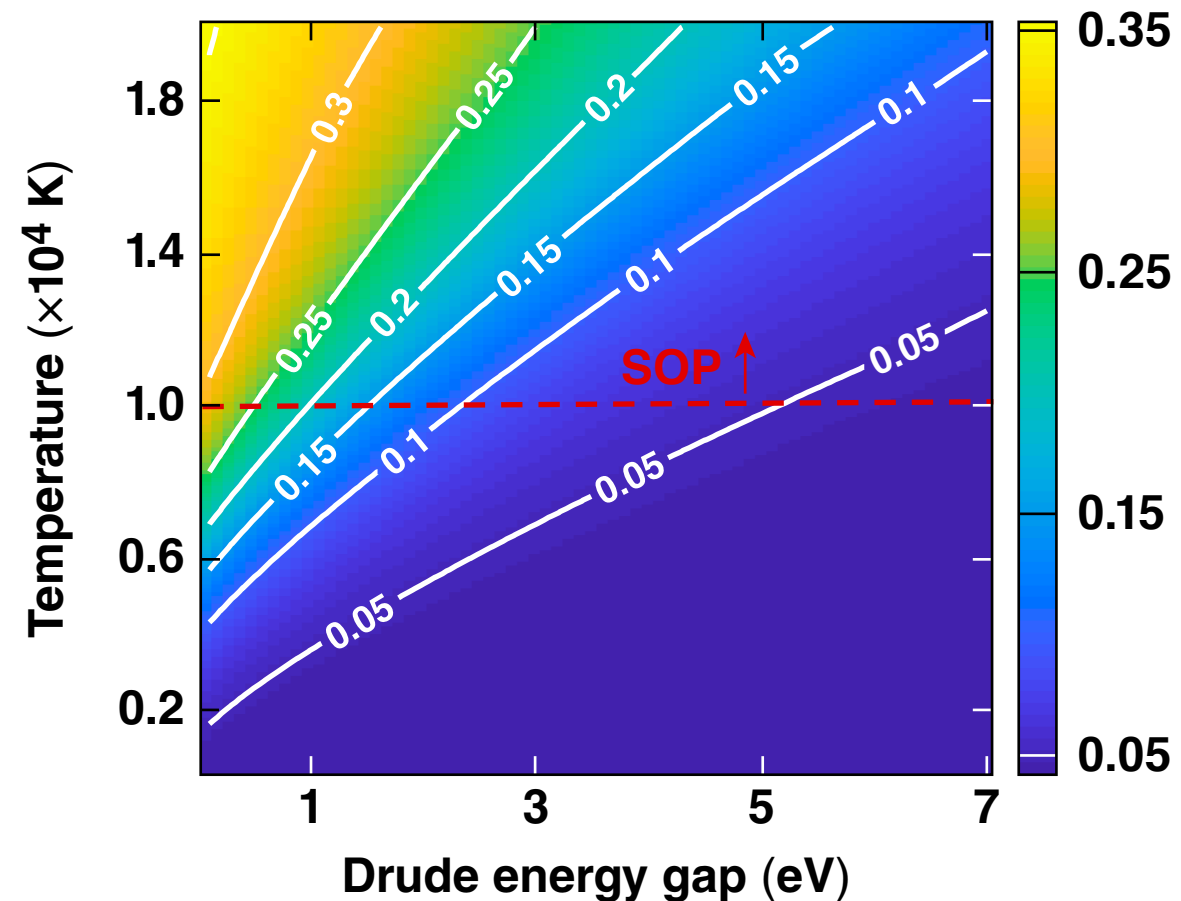
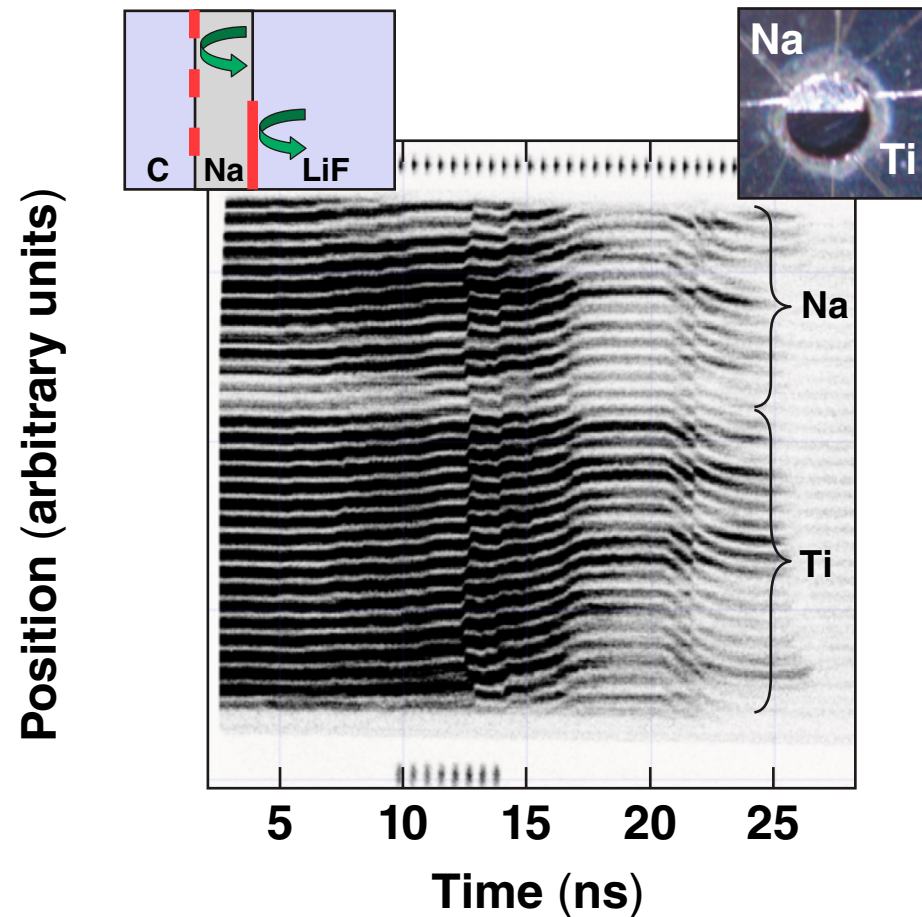
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- High-power lasers ramp compressed Na and nanosecond *in-situ* x-ray diffraction (XRD) was used to measure the crystal structure at pressures up to 418 GPa
- Solid Na was observed at 301 to 418 GPa; therefore, the melting temperature increases dramatically above 140 GPa
- Simultaneous XRD and optical reflectivity measurements reveal that Na remains reflective to at least 290 GPa, in striking contrast to room-temperature static-compression measurements

# Na remains reflective to at least 290 GPa, and the variation in the reflectivity is caused by changes in the window



Elevated temperatures may cause delocalization of the electrons and thermal population of the conduction band in Na.

# Reflectivity measurements of Na suggest a gap has opened to 2 eV at 290 GPa



Elevated temperatures may cause delocalization of the electrons and thermal population of the conduction band in Na.